

Today's Date: 6/13/2001

DB Name	Query	Hit Count	Set Name
USPT,PGPB,JPAB,EPAB,DWPI	11 and 12 and 113	7	<u>L30</u>
USPT,PGPB,JPAB,EPAB,DWPI	gauglitz\$.in. and 11	1	<u>L29</u>
USPT,PGPB,JPAB,EPAB,DWPI	steinwand\$.in. and 11	1	<u>L28</u>
USPT,PGPB,JPAB,EPAB,DWPI	brecht\$.in. and 11	1	<u>L27</u>
USPT,PGPB,JPAB,EPAB,DWPI	stemmler\$.in. and 11	1	<u>L26</u>
USPT,PGPB,JPAB,EPAB,DWPI	122 and 11	23	<u>L25</u>
USPT,PGPB,JPAB,EPAB,DWPI	((fluoresc\$5 adj2 quench\$3) near10 ((solid adj1 phase\$1) or microwell\$1 or microtiter\$1 or well\$1)) and 1	28	<u>L24</u>
USPT,PGPB,JPAB,EPAB,DWPI	12 and 122	15	<u>L23</u>
USPT,PGPB,JPAB,EPAB,DWPI	(fluoresc\$5 adj2 quench\$3) near10 ((solid adj1 phase\$1) or microwell\$1 or microtiter\$1 or well\$1)	47	<u>L22</u>
USPT,PGPB,JPAB,EPAB,DWPI	(fluoresc\$5 adj2 quench\$3) and 113 and 111 and 12	5	<u>L21</u>
USPT,PGPB,JPAB,EPAB,DWPI	((fluoresc\$5 adj2 quench\$3) near5 (solid adj1 phase\$1)) and l11 and l2	0	<u>L20</u>
USPT,PGPB,JPAB,EPAB,DWPI	(fluoresc\$5 adj2 quench\$3) near5 (coat\$3) near5 (solid adj1 phase\$1)	0	<u>L19</u>
USPT,PGPB,JPAB,EPAB,DWPI	(fluoresc\$5 adj2 quench\$3) near3 (coat\$3) near3 (solid adj1 phase\$1)	0	<u>L18</u>
USPT,PGPB,JPAB,EPAB,DWPI	113 and 11	34	<u>L17</u>
USPT,PGPB,JPAB,EPAB,DWPI	113 and 19 and 11	7	<u>L16</u>
USPT,PGPB,JPAB,EPAB,DWPI	112 and 113	4	<u>L15</u>
USPT,PGPB,JPAB,EPAB,DWPI	110 and 113	4	<u>L14</u>
USPT,PGPB,JPAB,EPAB,DWPI	(solid adj1 phase\$1) same (microtiter or nanotiter or microwell\$1 or well\$3) same quench\$3	55	<u>L13</u>
USPT,PGPB,JPAB,EPAB,DWPI	110 and 111	207	<u>L12</u>
USPT,PGPB,JPAB,EPAB,DWPI	(solid adj1 phase\$1) same (microtiter or nanotiter or microwell\$1 or well\$3)	9203	<u>L11</u>
USPT,PGPB,JPAB,EPAB,DWPI	19 and 18	320	<u>L10</u>
USPT,PGPB,JPAB,EPAB,DWPI	phase\$1 near2 (different or separat\$3)	129439	<u>L9</u>
USPT,PGPB,JPAB,EPAB,DWPI	((solid and liquid) near5 phase\$1) and 17	417	<u>L8</u>
USPT,PGPB,JPAB,EPAB,DWPI	15 and 16	1205	<u>L7</u>
USPT,PGPB,JPAB,EPAB,DWPI	phase\$1 near10 (assay\$3 or immunoassay\$1)	7556	<u>L6</u>
USPT,PGPB,JPAB,EPAB,DWPI	(affinity or immunoaffinity or competitive or sandwich) and 14	3623	<u>L5</u>
USPT,PGPB,JPAB,EPAB,DWPI	((quantitat\$3 or qualitat\$3) or ((interact\$3 or react\$3) adj3 kinetic\$1)) and l3	5826	<u>L4</u>
USPT,PGPB,JPAB,EPAB,DWPI	11 and 12	14880	<u>L3</u>
USPT,PGPB,JPAB,EPAB,DWPI	phase\$1 near5 (different or separat\$3)	169789	<u>L2</u>
USPT,PGPB,JPAB,EPAB,DWPI	phase\$1 and (assay\$3 or immunoassay\$1)	55311	<u>L1</u>

	A series and the series of the		The state of the s
	No. of the state o		
		,	•
			·
	4 Ev V		
*		4.3	
Since of the second			
₽.	*		
111			, f.
oj *			- 1
· ·			
4			
	₩		
			÷
* ;	•		
Sales Control			
7			
*			
- A			
÷			
			*
**************************************			
· 3			
A CA			
Ψ.			
			2.00
- 54			· 1
18. 19.			Ž.
Š			
100 Mg		The state of the s	இர். இத்தை இது
Ĺ			
*			
4			
			y van
-			
in.			
₽°			
÷			
<i>Y</i> .	Service Control of the Control of th		
4 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1			
(1) 12			
			- 150 m
		3(of 3	
A Rose			

```
FILE 'MEDLINE, EMBASE, SCISEARCH, CAPLUS, BIOSIS' ENTERED AT 15:27:05 ON
     13 JUN 2001
         83186 S PHASE? (3A) (SEPARATION OR DIFFERENT)
L1
L2
         181586 S PHASE (3A) (SOLID)
        147290 S PHASE (3A) (LIQUID)
L3
        163019 S PHASE? (3A) (LIQUID)
L4
L5
        198135 S PHASE? (3A) (SOLID)
            656 S L1 AND L2 AND L3
L6
L7
            126 S L6 AND ?ASSAY?
              0 S L7 AND (FLUORESCENCE QUENCH?)
L8
              0 S L7 AND (FLUORESCENCE (5P) QUENCH?)
L9
L10
          25653 S (FLUORESCENCE QUENCH?)
L11
            125 S L10 AND L1
L12
             3 S L11 AND L5
             14 S L11 AND L4
L13
             2 DUP REM L12 (1 DUPLICATE REMOVED)
L14
L15
             10 DUP REM L13 (4 DUPLICATES REMOVED)
             63 DUP REM L11 (62 DUPLICATES REMOVED)
L16
            59 DUP REM L7 (67 DUPLICATES REMOVED)
L17
            41 S L7 AND (INTERACT? OR REACT? OR KINETIC?)
L18
             18 S L7 AND FLUORESCENCE
L19
             0 S L7 AND QUENCH?
9 DUP REM L19 (9 DUPLICATES REMOVED)
L20
L21
             19 DUP REM L18 (22 DUPLICATES REMOVED)
L22
             0 S L18 AND (MICROTITER? OR MICROWELL?)
L23
             0 S L7 AND (MICROTITER? OR MICROWELL?)
L24
L25
             0 S L6 AND (MICROTITER? OR MICROWELL?)
            34 S L1 AND (MICROTITER? OR MICROWELL?)
L26
            17 DUP REM L26 (17 DUPLICATES REMOVED)
L27
```

1 S L27 AND FLUORESCENCE

L28

L14 ANSWER 2 OF 2 MEDLINE DUPLICATE 1

ACCESSION NUMBER: 85125284 MEDLINE

DOCUMENT NUMBER: 85125284 PubMed ID: 3882272

TITLE: Fluoroimmunoassays and immunofluorometric assays.

AUTHOR: Hemmila I

SOURCE: CLINICAL CHEMISTRY, (1985 Mar) 31 (3) 359-70. Ref: 124

Journal code: DBZ; 9421549. ISSN: 0009-9147.

PUB. COUNTRY: United States

Journal; Article; (JOURNAL ARTICLE)

General Review; (REVIEW)

LANGUAGE: English

FILE SEGMENT: Priority Journals

ENTRY MONTH: 198504

ENTRY DATE: Entered STN: 19900320

Last Updated on STN: 19900320 Entered Medline: 19850403

AB Fluorescent probes and fluorometric methods have gained increasing interest in the field of clinical immunology, not only as one additional alternative to radioimmunoassays, but also in producing cheap, stable,

and

safe reagents and rapid and sensitive assays. One of the main goals has been the development of homogeneous assays: assays based on fluorescence polarization, fluorescence quenching, excitation transfer, or enzymically releasable probes are widely applied, especially in drug monitoring. The development of suitable solid-phase separation techniques has facilitated utilization

of fluorescence in heterogeneous assays, which in general have wider applications, from proteins and viruses to small haptens. Lately new alternative fluorescent probes and methods have been introduced. For example, the use of fluorescent phycobil-iproteins or porphyrin

derivatives

with long-wavelength emission and large Stokes shift or, in particular, the rare earth chelates with unique fluorescent properties well suited to time-resolved measurement have opened new possibilities towards more sensitive immunoassays.

AB . . . and sensitive assays. One of the main goals has been the development of homogeneous assays: assays based on fluorescence polarization, fluorescence quenching, excitation transfer, or enzymically releasable probes are widely applied, especially in drug monitoring. The development of suitable solid-phase separation techniques has facilitated utilization of fluorescence in heterogeneous assays, which in general have wider applications, from proteins and viruses to. . .

L21 ANSWER 1 OF 9 SCISEARCH COPYRIGHT 2001 ISI (R) DUPLICATE 1

ACCESSION NUMBER: 1999:142267 SCISEARCH

THE GENUINE ARTICLE: 165PR

TITLE: Selective trace enrichment by immunoaffinity capillary

electrochromatography on-line with capillary zone

electrophoresis - laser-induced fluorescence

AUTHOR: Thomas D H; Rakestraw D J; Schoeniger J S (Reprint);

LopezAvila V; VanEmon J

CORPORATE SOURCE: SANDIA NATL LABS, POB 969 MS 9671, LIVERMORE, CA 94551

(Reprint); SANDIA NATL LABS, LIVERMORE, CA 94551; MIDWEST

RES INST, CALIF OPERAT, MT VIEW, CA; US EPA, NATL

**EXPOSURE** 

DOCUMENT TYPE:

RES LAB, HUMAN EXPOSURE RES BRANCH, LAS VEGAS, NV 89193

COUNTRY OF AUTHOR: USA

SOURCE: ELECTROPHORESIS, (JAN 1999) Vol. 20, No. 1, pp. 57-66.

Publisher: WILEY-V C H VERLAG GMBH, MUHLENSTRASSE 33-34,

D-13187 BERLIN, GERMANY.

ISSN: 0173-0835. Article; Journal

FILE SEGMENT: LIFE LANGUAGE: English

REFERENCE COUNT: 30

\*ABSTRACT IS AVAILABLE IN THE ALL AND IALL FORMATS\*

AB Limited by the lack of a sensitive, universal detector, many

capillary-based liquid-phase separation

techniques might benefit from techniques that overcome modest

concentration sensitivity by preconcentrating large injection volumes.

The

work presented employs selective solid-phase

extraction by immunoaffinity capillary electrochromatography (IACEC) to enhance detection limits. A model analyte, fluorescein isothiocyanate (FITC) biotin, is electrokinetically applied to a capillary column packed with an immobilized anti-biotin-IgG support. After selective extraction

by

the immunoaffinity capillary, the bound analyte is eluted, migrates by capillary zone electrophoresis (CZE), and is detected by laser-induced fluorescence. The column is regenerated and reused many times. We evaluate the performance of IACEC for selective trace enrichment of analytes prior to CZE. The calibration curve for FITC-biotin bound versus application time is linear from 10 to 300 seconds. Recovery of FITC-biotin

spiked into a diluted urinary metabolites solution was 89.4% Versus spiked

buffer, with a precision of 1.8% relative standard deviation (RSD).

TI Selective trace enrichment by immunoaffinity capillary electrochromatography on-line with capillary zone electrophoresis - laser-induced **fluorescence** 

AB Limited by the lack of a sensitive, universal detector, many capillary-based liquid-phase separation techniques might benefit from techniques that overcome modest concentration sensitivity by preconcentrating large injection volumes.

The

work presented employs selective **solid-phase** extraction by immunoaffinity capillary electrochromatography (IACEC) to enhance detection limits. A model analyte, fluorescein isothiocyanate (FITC) biotin, is electrokinetically applied. . . by the

immunoaffinity

capillary, the bound analyte is eluted, migrates by capillary zone electrophoresis (CZE), and is detected by laser-induced

fluorescence. The column is regenerated and reused many times. We evaluate the performance of IACEC for selective trace enrichment of analytes.

STP KeyWords Plus (R): AFFINITY-CHROMATOGRAPHY; ISOTACHOPHORESIS;

PRECONCENTRATION; PRETREATMENT; IMMUNOASSAY; BINDING

L21 ANSWER 2 OF 9 EMBASE COPYRIGHT 2001 ELSEVIER SCI. B.V.

ACCESSION NUMBER: 97086383 EMBASE

DOCUMENT NUMBER: 1997086383

TITLE: Quantitation of an orally available thrombin inhibitor in

rat, monkey and human plasma and in human urine by high-performance liquid chromatography and fluorescent

post-column derivatization of arginine.

AUTHOR: Mendoza C.B.; Dixon S.A.; Lods M.M.; Ma M.G.; Nguyen K.T.;

Nutt R.F.; Tran H.S.; Nolan T.G.

CORPORATE SOURCE: T.G. Nolan, Corvas International, Department of Analytical

Chemistry, 3030 Science Park Road, San Diego, CA

92121-1102, United States

SOURCE: Journal of Chromatography A, (1997) 762/1-2 (299-310).

Refs: 16

ISSN: 0021-9673 CODEN: JCRAEY

PUBLISHER IDENT.: S 0021-9673(96)00865-5

COUNTRY: Netherlands

DOCUMENT TYPE: Journal; Conference Article

FILE SEGMENT: 025 Hematology 030 Pharmacology

037 Drug Literature Index

LANGUAGE: English SUMMARY LANGUAGE: English

AB An **assay** for the quantification of plasma and urine levels of CVS 1123, an orally bioavailable thrombin inhibitor, and its desmethyl form, CVS 738, was developed to support clinical and toxicology studies.

This assay uses solid-phase extraction,

reversed-phase HPLC separation, and post-column fluorescent derivatization with ninhydrin. An internal standard is added to correct for recovery. In aqueous solution, the arginine aldehyde structures of CVS 1123 and CVS 738 exist in multiple forms which can be separated under standard reversed-phase HPLC conditions. HPLC conditions were optimized to give rapid interconversion of the forms on the separation time scale, and consequently a single chromatographic peak. Extraction conditions were modified for quantitative extraction of drug compounds from large volumes of human plasma. The assay was shown to be accurate and precise, with a quantification limit of 17 ng

CVS

1123/ml human plasma.

AB An assay for the quantification of plasma and urine levels of CVS 1123, an orally bioavailable thrombin inhibitor, and its desmethyl form, CVS 738, was developed to support clinical and toxicology studies. This assay uses solid-phase extraction,

reversed-phase HPLC separation, and post-column

fluorescent derivatization with ninhydrin. An internal standard is added to correct for recovery. In aqueous solution, the arginine. . . single chromatographic peak. Extraction conditions were modified for quantitative

extraction of drug compounds from large volumes of human plasma. The assay was shown to be accurate and precise, with a quantification limit of 17 ng CVS 1123/ml human plasma.

CT Medical Descriptors:

\*drug blood level \*drug urine level

accuracy
animal tissue
conference paper
controlled study
derivatization

drug determination

```
high performance liquid chromatography
     human tissue
     monkey
     nonhuman
     priority journal
     quantitative assay
     reversed phase high performance liquid chromatography
     solid phase extraction
     technique
     *cvs 1123: AN, drug analysis
     *cvs 1123: CR, drug concentration
     *thrombin inhibitor: AN, drug analysis
     *thrombin inhibitor: CR, drug concentration
     unclassified drug
L21 ANSWER 3 OF 9 MEDLINE
                                                        DUPLICATE 2
ACCESSION NUMBER:
                    96285926
                                 MEDLINE
DOCUMENT NUMBER:
                    96285926
                             PubMed ID: 8704933
                    Liquid chromatographic assay for a butenolide
TITLE:
                    endothelin antagonist (PD 156707) in plasma.
AUTHOR:
                    Rossi D T; Hallak H; Bradford L
CORPORATE SOURCE:
                    Department of Pharmacokinetics and Drug Metabolism,
                    Parke-Davis Pharmaceutical Research, Division of Warner
                    Lambert Company, Ann Arbor, MI 48105, USA.
SOURCE:
                    JOURNAL OF CHROMATOGRAPHY. B, BIOMEDICAL APPLICATIONS,
                    (1996 Mar 3) 677 (2) 299-304.
                    Journal code: BXL; 9421796. ISSN: 0378-4347.
PUB. COUNTRY:
                    Netherlands
                    Journal; Article; (JOURNAL ARTICLE)
                    English
LANGUAGE:
FILE SEGMENT:
                    Priority Journals
ENTRY MONTH:
                    199609
ENTRY DATE:
                    Entered STN: 19960919
                    Last Updated on STN: 19960919
                    Entered Medline: 19960912
    A sensitive and selective liquid chromatographic assay for
    determining the non-peptide endothelin A receptor antagonist PD 156707
(I)
     in rat plasma has been developed and validated. The analyte was isolated
     from matrix by solid-phase extraction. Liquid
     chromatographic separation was achieved isocratically on a 3.2
    mm I.D., ODS column with a mobile phase of acetonitrile-ammonium
phosphate
     (50 mM, pH 3.5) (44:56, v/v). Column effluent was monitored
     fluorometrically. Peak-height ratios (analyte/IS) were proportional to I
    concentrations in rat plasma from 25 to 1000 ng/ml. Assay
    precision and accuracy for I, based on quality controls, was 9.5%
relative
     standard deviation, with relative error of \pm 1 6.5%. The quantitation
     limit was 25 ng/ml for a 200-microliters sample aliquot.
    Liquid chromatographic assay for a butenolide endothelin
    antagonist (PD 156707) in plasma.
    A sensitive and selective liquid chromatographic assay for
    determining the non-peptide endothelin A receptor antagonist PD 156707
(I)
     in rat plasma has been developed and validated. The analyte was isolated
    from matrix by solid-phase extraction. Liquid
    chromatographic separation was achieved isocratically on a 3.2
    mm I.D., ODS column with a mobile phase of acetonitrile-ammonium
phosphate
     (50 mM, pH. . . effluent was monitored fluorometrically. Peak-height
    ratios (analyte/IS) were proportional to I concentrations in rat plasma
```

from 25 to 1000 ng/ml. Assay precision and accuracy for I, based

fluorescence

AB

TI

AB

```
on quality controls, was 9.5% relative standard deviation, with relative
     error of +/- 6.5%..
CT
Pressure Liquid: MT, methods
     *Dioxoles: BL, blood
      Rats
     *Receptors, Endothelin: AI, antagonists & inhibitors
      Reproducibility of Results
      Sensitivity and Specificity
      Spectrometry, Fluorescence
L21 ANSWER 4 OF 9 EMBASE COPYRIGHT 2001 ELSEVIER SCI. B.V.
                    96166988 EMBASE
ACCESSION NUMBER:
DOCUMENT NUMBER:
                    1996166988
TITLE:
                    Determination of alosetron in human plasma or serum by
                    high-performance liquid chromatography with robotic sample
                    preparation.
AUTHOR:
                    Lloyd T.L.; Gupta S.K.; Gooding A.E.; Alianti J.R.
CORPORATE SOURCE:
                    Glaxo Research Institute, 5 Moore Drive, Research Triangle
                    Park, NC 27709, United States
SOURCE:
                    Journal of Chromatography B: Biomedical Applications,
                    (1996) 678/2 (261-267).
                    ISSN: 0378-4347 CODEN: JCBBEP
COUNTRY:
                    Netherlands
DOCUMENT TYPE:
                    Journal; Article
FILE SEGMENT:
                    027
                            Biophysics, Bioengineering and Medical
                            Instrumentation
                    029
                            Clinical Biochemistry
                    032
                            Psychiatry
                    048
                            Gastroenterology
                    030
                            Pharmacology
                            Drug Literature Index
                    037
LANGUAGE:
                    English
SUMMARY LANGUAGE:
                    English
AB
     A method of analysis for the determination of alosetron in human plasma
or
     serum has been developed. The method was fully automated using a
     laboratory robot in order to improve analytical precision, efficiency and
     safety. The assay involved solid-phase
     extraction with reversed-phase HPLC separation and
     fluorescence detection. A validation exercise over the
     concentration range of 0.1 to 20 ng/ml demonstrated the selectivity,
     linearity, sensitivity, accuracy, precision, extraction efficiency,
     ruggedness and stability of the method. The method has been applied in
     support of numerous human pharmacokinetic/biopharmaceutic studies over
the
     last five years.
     . . developed. The method was fully automated using a laboratory
AB
     robot in order to improve analytical precision, efficiency and safety.
The
     assay involved solid-phase extraction with
     reversed-phase HPLC separation and
     fluorescence detection. A validation exercise over the
     concentration range of 0.1 to 20 ng/ml demonstrated the selectivity,
     linearity, sensitivity, accuracy, precision,. . .
CT
    Medical Descriptors:
     *drug blood level
     *drug determination
     article
     controlled study
     fluorescence
     high performance liquid chromatography
    human experiment
```

human tissue

oral drug administration

priority journal

reversed phase high performance liquid chromatography robotics

## solid phase extraction

technique

\*alosetron: AN, drug analysis \*alosetron: PK, pharmacokinetics \*alosetron: CR, drug concentration

L21 ANSWER 5 OF 9 MEDLINE DUPLICATE 3

ACCESSION NUMBER: 96043270

MEDLINE DOCUMENT NUMBER: 96043270

PubMed ID: 7496992

Mammalian secreted and cytosolic phospholipase A2 show TITLE:

different specificities for phospholipid molecular

species.

Burdge G C; Creaney A; Postle A D; Wilton D C AUTHOR:

Department of Biochemistry, University of Southampton, CORPORATE SOURCE:

U.K.

SOURCE: INTERNATIONAL JOURNAL OF BIOCHEMISTRY AND CELL BIOLOGY,

(1995 Oct) 27 (10) 1027-32.

Journal code: CDK; 9508482. ISSN: 1357-2725.

PUB. COUNTRY:

ENGLAND: United Kingdom

Journal; Article; (JOURNAL ARTICLE)

LANGUAGE: English

FILE SEGMENT: Priority Journals

ENTRY MONTH: 199601

ENTRY DATE: Entered STN: 19960217

> Last Updated on STN: 19960217 Entered Medline: 19960117

AB Previous studies using phospholipid vesicles containing single molecular species have shown cytosolic phospholipase (85 kDa) (PL) A2 to possess a marked preference for arachidonic acid (20:4n-6)-containing species,

secreted PLA2 (14 kDa) exhibited little acyl chain selectivity. In this study, we have defined the molecular specificity of cytosolic PLA2 using phospholipid vesicles derived from rat liver which contain complex mixtures of molecular species. Phosphatidylcholine (PC) and phosphatidylethanolamine (PE) were isolated from rat liver by chloroform and methanol extraction, and solid-phase

separation. PC and PE vesicles were hydrolysed by either human recombinant cytosolic or porcine pancreatic PLA2. Molecular species compositions were determined by reverse phase high performance liquid chromatography (HPLC) with post-column fluorescence derivitisation. HPLC analysis after limited hydrolysis demonstrated that the secreted phospholipase A2 showed no significant acyl chain

specificity using these phospholipid mixtures. However, the cytosolic enzyme demonstrated a high degree of preference for arachidonic acid-containing

species such that there was no hydrolysis of other molecular species. The extent of hydrolysis of PC16:0/20:4 was 1.4-fold greater (P < 0.05, n =

3)

confirms

than PC18:0/20:4, while PE16:0/20:4 and PE18:0/20:4 were hydrolysed to a similar degree. Under these assay conditions, the cytosolic enzyme showed a preference for PE as compared with PC. This study

that cytosolic PLA2 is highly selective for sn-2 20:4n-6-containing phospholipid molecular species even when presented with a complex natural species mixture. This specificity is consistent with the cytosolic enzyme having a primary role in the process of arachidonic release within cells. (ABSTRACT TRUNCATED AT 250 WORDS)

AB . . . mixtures of molecular species. Phosphatidylcholine (PC) and phosphatidylethanolamine (PE) were isolated from rat liver by chloroform and methanol extraction, and solid-phase

separation. PC and PE vesicles were hydrolysed by either human recombinant cytosolic or porcine pancreatic PLA2. Molecular species compositions were determined by reverse phase high performance

liquid chromatography (HPLC) with post-column fluorescence
 derivitisation. HPLC analysis after limited hydrolysis demonstrated that
 the secreted phospholipase A2 showed no significant acyl chain
specificity

using these. . . (P < 0.05, n = 3) than PC18:0/20:4, while PE16:0/20:4 and PE18:0/20:4 were hydrolysed to a similar degree. Under these assay conditions, the cytosolic enzyme showed a preference for PE as compared with PC. This study confirms that cytosolic PLA2 is. . .

L21 ANSWER 6 OF 9 MEDLINE

DUPLICATE 4

ACCESSION NUMBER:

95005369 MEDLINE

DOCUMENT NUMBER:

95005369 PubMed ID: 7921173

TITLE:

Determination of a novel hematoregulatory peptide in dog

plasma by reversed-phase high-performance
liquid chromatography and an amine-selective

o-phthaldialdehyde-thiol post-column reaction with

fluorescence detection.

AUTHOR:

Boppana V K; Miller-Stein C

CORPORATE SOURCE:

Department of Drug Metabolism and Pharmacokinetics, SmithKline Beecham Pharmaceuticals, King of Prussia,

PA-19406.

SOURCE:

JOURNAL OF CHROMATOGRAPHY. A, (1994 Jul 29) 676 (1) 161-7.

Journal code: BXJ; 9318488.

PUB. COUNTRY:

Netherlands

Journal; Article; (JOURNAL ARTICLE)

LANGUAGE:

English

FILE SEGMENT:

Priority Journals

ENTRY MONTH:

199410

ENTRY DATE:

Entered STN: 19941222

Last Updated on STN: 19941222 Entered Medline: 19941025

A sensitive and selective high-performance liquid chromatographic method AB was developed for the determination of SB 107647 (I), a novel synthetic hematoregulatory peptide, in plasma samples of dog and rat. The method involves isolation of I and the internal standard (SB 203285, IS) from plasma by a solid-phase anion-exchange extraction column prior to reversed-phase ion-pair chromatographic separation on an octyl silica column. Following separation, a selective post-column reaction of the epsilon-amino groups of the lysine moieties of the peptide with o-phthaldialdehyde and a thiol under basic conditions was used to generate a highly fluorescent isoindole product, which was subsequently detected on-line with a fluorometer. Optimization of chromatographic conditions resulted in an on-column detection limit of 1 ng. The recovery of I from dog plasma at 20 and 4000 ng/ml was 50.0 +/-5.94 and 56.6 +/-1.45% (Mean +/-S.D.), respectively. The limit of quantification for I, for 0.25-ml plasma samples, was 20 ng/ml. Linear response was observed for concentrations of I ranging from 20 to 4000 ng/ml of plasma. The assay was sufficiently sensitive, accurate and precise to support toxicokinetic studies in animal species. TI Determination of a novel hematoregulatory peptide in dog plasma by reversed-phase high-performance liquid chromatography and an amine-selective o-phthaldialdehyde-thiol post-column reaction with

fluorescence detection.

AB . . . dog and rat. The method involves isolation of I and the internal standard (SB 203285, IS) from plasma by a solid-phase anion-exchange extraction column prior to reversed-phase ion-pair chromatographic separation on an octyl silica column. Following separation, a selective post-column reaction of the epsilon-amino groups of the lysine moieties of. . . was 20 ng/ml. Linear response was observed for concentrations of I ranging from 20 to 4000 ng/ml of plasma. The assay was sufficiently sensitive, accurate and precise to support toxicokinetic studies in animal species.

L21 ANSWER 7 OF 9 EMBASE COPYRIGHT 2001 ELSEVIER SCI. B.V.

ACCESSION NUMBER:

93029264 EMBASE

DOCUMENT NUMBER:

1993029264

TITLE: Method for the determination of indole-3-acetic acid and

related compounds of L-tryptophan catabolism in soils.

AUTHOR: Lebuhn M.; Hartmann A.

CORPORATE SOURCE:

GSF, Forsch.-zent. Umwelt/Gesundheit GmbH, Institut fur

Bodenokologie, Ingolstadter Landstrasse 1, W-8042

Neuherberg, Germany

SOURCE:

Journal of Chromatography, (1993) 629/2 (255-266).

ISSN: 0021-9673 CODEN: JOCRAM

COUNTRY:

Netherlands

DOCUMENT TYPE:

Journal; Article

FILE SEGMENT:

029 Clinical Biochemistry

046 Environmental Health and Pollution Control

LANGUAGE:

English

SUMMARY LANGUAGE:

English

An optimized method for the determination of substances occurring in auxin

metabolism and L-tryptophan (TRP) catabolism was developed. It is based on

solid-phase extraction (SPE), two isocratic reversedphase high-performance liquid chromatographic (HPLC)

separations at different liquid phase

conditions and the simultaneous detection of fluorescence and UV absorbance at different wavelengths. Advantages of the proposed method are: the solvent (ethanol) and liquid phase

(containing 2-propanol) provide optimum stability and selectivity; almost no toxic wastes are produced; no time-consuming liquid-liquid extractions (LLE), derivatization procedures or column re-equilibration (obligatory for gradient systems) are necessary, no need for antioxidants, ion-pair

or

derivatization reagents; recovery rates of the SPE system are superior to LLE effeciencies; high sensitivity, selectivity and identification capacity are provided by the proposed HPLC and detection system. By measuring various chromatographic and spectral parameters simultaneously, the determination reliability is improved. The characteristic chromatographic and spectral data for selected indole derivatives and TRP catabolites are presented. In samples from two different soils that were tested with the proposed method, the actual contents of TRP were 1.4 and 5.8 .mu.q/q dry soil. In addition, traces of indole-3-acetic acid (IAA) could be detected. When TRP was added, IAA was the predominant catabolite in both soils, and reached values of 2.9 and 8.0 .mu.g/g dry soil. In addition to IAA, indole-3-ethanol, indole-3-aldehyde, indole-3-carboxylic acid, indole-3-lactic acid, anthranilic acid and traces of indole-3-acetamide were identified and determined.

. . . method for the determination of substances occurring in auxin AB metabolism and L-tryptophan (TRP) catabolism was developed. It is based on

solid-phase extraction (SPE), two isocratic reversedphase high-performance liquid chromatographic (HPLC)

separations at different liquid phase

conditions and the simultaneous detection of fluorescence and UV absorbance at different wavelengths. Advantages of the proposed method are: the solvent (ethanol) and liquid phase

(containing 2-propanol) provide optimum stability and selectivity; almost no toxic wastes are produced; no time-consuming liquid-liquid extractions (LLE), derivatization procedures. . .

Medical Descriptors:

\*assav

article catabolism priority journal soil \*indoleacetic acid \*tryptophan

L21 ANSWER 8 OF 9 MEDLINE ACCESSION NUMBER: 88115776 DUPLICATE 5

MEDLINE

DOCUMENT NUMBER: 88115776 PubMed ID: 3429585

Liquid chromatographic determination of sotalol in plasma

and urine employing solid-phase

extraction and fluorescence detection.

AUTHOR: Bartek M J; Vekshteyn M; Boarman M P; Gallo D G

CORPORATE SOURCE: Department of Metabolism and Pharmacokinetics,

Bristol-Myers Company, Evansville, IN 47721. JOURNAL OF CHROMATOGRAPHY, (1987 Oct 30) 421 (2) 309-18. SOURCE:

Journal code: HQF; 0427043. ISSN: 0021-9673.

PUB. COUNTRY: Netherlands

Journal; Article; (JOURNAL ARTICLE)

LANGUAGE: English

FILE SEGMENT: Priority Journals

ENTRY MONTH: 198803

ENTRY DATE: Entered STN: 19900308

Last Updated on STN: 19900308 Entered Medline: 19880322

AB A liquid chromatographic method using a solid-phase

extraction procedure for the quantification of sotalol in plasma and urine

is described. Sotalol is eluted from an extraction column with ethyl acetate-acetonitrile (1:2) and, after separation by reversedphase high-performance liquid chromatography on a mu Bondapak C18 column, is quantified by fluorescence detection at excitation and emission wavelengths of 240 and 310 nm, respectively. The method has been demonstrated to be linear over the concentration ranges

10-6000 ng/ml in plasma and 0.5-100 micrograms/ml in urine. Mean interassay accuracy of the method for plasma ranged from 93 to 100% and for urine from 102 to 114%; precision ranged from 0.5 to 1.6% for plasma over a concentration range of 200-4000 ng/ml and for urine from 0.7 to 2.0% at concentrations of 2-50 micrograms/ml. Mass spectrometry confirmed the presence of sotalol in isolated chromatographic fractions of plasma and urine extracts from subjects given sotalol orally.

Liquid chromatographic determination of sotalol in plasma and urine ΤI employing solid-phase extraction and fluorescence detection.

A liquid chromatographic method using a solid-phase ΑB extraction procedure for the quantification of sotalol in plasma and urine

is described. Sotalol is eluted from an extraction column with ethyl acetate-acetonitrile (1:2) and, after separation by reversedphase high-performance liquid chromatography on a mu Bondapak C18 column, is quantified by fluorescence detection at excitation and emission wavelengths of 240 and 310 nm, respectively. The

method has been demonstrated to be linear over the concentration ranges 10-6000 ng/ml in plasma and 0.5-100 micrograms/ml in urine. Mean interassay accuracy of the method for plasma ranged from 93 to 100% and for urine from 102 to 114%; precision ranged.

Human CT

Chromatography, High Pressure Liquid

Chromatography, Liquid

Drug Stability

Mass Fragmentography \*Sotalol: AN, analysis Sotalol: BL, blood Sotalol: UR, urine

Spectrometry, Fluorescence

L21 ANSWER 9 OF 9 MEDLINE

ACCESSION NUMBER: 86304791 MEDLINE

PubMed ID: 3745383 DOCUMENT NUMBER: 86304791 TITLE: Solid-phase extraction and

determination of dansyl derivatives of unconjugated and

acetylated polyamines by reversed-phase liquid chromatography: improved separation

systems for polyamines in cerebrospinal fluid, urine and

tissue.

AUTHOR: Kabra P M; Lee H K; Lubich W P; Marton L J

CONTRACT NUMBER: CA-13525 (NCI)

CA-37606 (NCI)

SOURCE: JOURNAL OF CHROMATOGRAPHY, (1986 Jul 11) 380 (1) 19-32.

Journal code: HQF; 0427043. ISSN: 0021-9673.

PUB. COUNTRY: Netherlands

Journal; Article; (JOURNAL ARTICLE)

LANGUAGE: English

FILE SEGMENT: Priority Journals

ENTRY MONTH: 198610

ENTRY DATE: Entered STN: 19900321

Last Updated on STN: 19970203 Entered Medline: 19861002

AB A sensitive and simple liquid chromatographic assay with fluorometric detection for unconjugated and acetylated polyamines in biological fluids is described. After precolumn derivatization with dansyl

chloride, unconjugated polyamines and acetylated polyamines were extracted

by elution from a Bond-Elut C18 column and then separated on a reversed-phase column with gradient elution. The complete analysis of unconjugated putrescine, spermidine, and spermine in either hydrolyzed urine, cerebrospinal fluid or tissue could be accomplished within 20-26 min, while the simultaneous analysis of unconjugated polyamines and monoacetylpolyamines could be completed within 40 min. Unhydrolyzed urine and cerebrospinal fluid required a Bond-Elut cation-exchange clean-up before dansylation. Standard curves for the assay were linear up to 20 nmol/ml, and the within-day and day-to-day coefficients of

variation were between 1.1 and 4.6% and between 1.6 and 11.8%, respectively. Results

obtained with the method were compared with results obtained with a well established modified amino acid analyzer method for urine, tissue and cerebrospinal fluid samples. The correlation coefficients between these two methods were in the range 0.933-0.996. Detection limits between 50

 $\,$  150 fmol were achieved for unconjugated and acetylated polyamines. Of more

than twenty drugs and amines tested for possible interference with the assay, only normetanephrine was found to have the same retention time as the internal standard 1,6-diaminohexane.

TI Solid-phase extraction and determination of dansyl derivatives of unconjugated and acetylated polyamines by reversed-phase liquid chromatography: improved separation systems for polyamines in cerebrospinal fluid, urine and tissue.

AB A sensitive and simple liquid chromatographic assay with fluorometric detection for unconjugated and acetylated polyamines in biological fluids is described. After precolumn derivatization with dansyl

chloride, unconjugated. . . completed within 40 min. Unhydrolyzed urine

and cerebrospinal fluid required a Bond-Elut cation-exchange clean-up before dansylation. Standard curves for the **assay** were linear up to 20 nmol/ml, and the within-day and day-to-day coefficients of variation

were between 1.1 and 4.6% and. . . were achieved for unconjugated and acetylated polyamines. Of more than twenty drugs and amines tested for possible interference with the assay, only normetanephrine was found to have the same retention time as the internal standard 1,6-diaminohexane.

High Pressure Liquid

CT

and

\*Dansyl Compounds: AN, analysis

Indicators and Reagents
\*Polyamines: AN, analysis

Polyamines: CF, cerebrospinal fluid Polyamines: UR, urine Spectrometry, Fluorescence

L22 ANSWER 7 OF 19 MEDLINE DUPLICATE 2

ACCESSION NUMBER: 97343989 MEDLINE

DOCUMENT NUMBER: 97343989 PubMed ID: 9200516

TITLE: High speed liquid chromatography of phenylethanolamines

for

the kinetic analysis of [110]-meta-

hydroxyephedrine and metabolites in plasma.

AUTHOR: Link J M; Synovec R E; Krohn K A; Caldwell J H

CORPORATE SOURCE: Department of Radiology, University of Washington, Seattle

98195-6004, USA.

CONTRACT NUMBER: HL 50238 (NHLBI) HL50239 (NHLBI)

SOURCE: JOURNAL OF CHROMATOGRAPHY. B, BIOMEDICAL SCIENCES AND

APPLICATIONS, (1997 May 23) 693 (1) 31-41.

Journal code: CXN; 9714109. ISSN: 1387-2273.

PUB. COUNTRY: Netherlands

Journal; Article; (JOURNAL ARTICLE)

LANGUAGE: English

FILE SEGMENT: Priority Journals

ENTRY MONTH: 199708

ENTRY DATE: Entered STN: 19970902

Last Updated on STN: 19980206

Entered Medline: 19970820

AB A method is developed and described for analysis of [11C]-meta-hydroxyephedrine, [11C]MHED, a tracer of cardiac function, and its metabolites in plasma samples. The method combines on-column solid -phase extraction and separation on a single weak cation-exchange column. Phenylethanolamines were used to develop the separation method that concentrates the analytes on-column from physiological saline and then elutes them by changing to an acidic mobile phase. Hydrophobic interactions determine the selectivity, and elution order is the same as for reversed-phase liquid chromatography on a Cl stationary phase. The mechanism of separation is mixed mode, with ion-exchange coupled with a reversed-phase liquid chromatography mechanism. Each sample analysis requires only 10 min and does not require deproteinization

or the use of organic solvents. In human samples, a single plasma metabolite of [11C]MHED along with the parent compound were observed using

this method. The method was sufficiently rapid so that in 70 min seven samples were **assayed**, providing a well-defined time course for MHED and its metabolites in blood. The metabolite concentration increased with time to approximately 85% of the plasma activity 50 min after administration. The results with the developed method are comparable to those described for reversed-phase separations, with the advantage that our method does not require deproteinization, reducing sample analysis time by a factor of two.

TI High speed liquid chromatography of phenylethanolamines for the **kinetic** analysis of [11C]-meta-hydroxyephedrine and metabolites in plasma.

AB . . . for analysis of [11C]-meta-hydroxyephedrine, [11C]MHED, a tracer of cardiac function, and its metabolites in plasma samples. The method combines on-column solid-phase extraction and

separation on a single weak cation-exchange column.

Phenylethanolamines were used to develop the separation method that concentrates the analytes on-column from physiological saline and then elutes them by changing to an acidic mobile phase. Hydrophobic

interactions determine the selectivity, and elution order is the
same as for reversed-phase liquid chromatography on a
Cl stationary phase. The mechanism of separation is
mixed mode, with ion-exchange coupled with a reversed-phase
liquid chromatography mechanism. Each sample analysis requires
only 10 min and does not require deproteinization or the use of organic
solvents.. . . parent compound were observed using this method. The
method was sufficiently rapid so that in 70 min seven samples were
assayed, providing a well-defined time course for MHED and its
metabolites in blood. The metabolite concentration increased with time to
approximately. . .

L22 ANSWER 8 OF 19 MEDLINE

DUPLICATE 3

ACCESSION NUMBER:

97084423

MEDLINE

DOCUMENT NUMBER:

97084423 PubMed ID: 8930766

TITLE:

Assay for taurine conjugates of bile acids in

serum by reversed-phase high-performance

liquid chromatography.

AUTHOR:

Paauw J D; Van Wyk L; Davis A T

CORPORATE SOURCE:

Department of Surgery, Michigan State University and

Butterworth Hospital, MI 495113, USA.

SOURCE:

JOURNAL OF CHROMATOGRAPHY. B, BIOMEDICAL APPLICATIONS,

(1996 Oct 11) 685 (1) 171-5.

Journal code: BXL; 9421796. ISSN: 0378-4347.

PUB. COUNTRY:

Netherlands

Journal; Article; (JOURNAL ARTICLE)

LANGUAGE:

English

FILE SEGMENT:

Priority Journals

ENTRY MONTH:

199702

ENTRY DATE:

Entered STN: 19970305

Last Updated on STN: 19970305 Entered Medline: 19970218

The purpose of this study was to develop a new high-performance liquid chromatographic (HPLC) procedure for quantifying taurine conjugates of bile acids in serum. The technique involved three basic steps. The first removed free amino acids via solid-phase extraction of the serum. The second step involved the reaction of the extracted serum with the enzyme choloylglycine hydrolase, which liberated the taurine from the conjugated bile acids. The third step was the reversed-phase HPLC separation of ophthalicdicarboxaldehyde derivatives of taurine. The assay provides a simple technique for determination of the total amount of taurine-conjugated bile acids in serum.

TI Assay for taurine conjugates of bile acids in serum by reversedphase high-performance liquid chromatography.

AB . . . taurine conjugates of bile acids in serum. The technique involved

three basic steps. The first removed free amino acids via **solid-phase** extraction of the serum. The second step involved the **reaction** of the extracted serum with the enzyme choloylglycine hydrolase, which liberated the taurine from the conjugated bile acids.

The

third step was the reversed-phase HPLC separation of o-phthalicdicarboxaldehyde derivatives of taurine. The assay provides a simple technique for determination of the total amount of taurine-conjugated bile acids in serum.

L22 ANSWER 14 OF 19 MEDLINE DUPLICATE 6

ACCESSION NUMBER: 91340933 MEDLINE

DOCUMENT NUMBER: 91340933 PubMed ID: 1874864
TITLE: Solid-phase extraction of plasma

vasopressin: evaluation, validation and application.

AUTHOR: Van de Heijning B J; Koekkoek-van den Herik I; Ivanyi T;

Van Wimersma Greidanus T B

CORPORATE SOURCE: Department of Pharmacology, Rudolf Magnus Institute,

University of Utrecht, The Netherlands.

SOURCE: JOURNAL OF CHROMATOGRAPHY, (1991 Apr 19) 565 (1-2) 159-71.

Journal code: HQF; 0427043. ISSN: 0021-9673.

PUB. COUNTRY: Netherlands

Journal; Article; (JOURNAL ARTICLE)

LANGUAGE: English

FILE SEGMENT: Priority Journals

ENTRY MONTH: 199109

ENTRY DATE: Entered STN: 19911013

Last Updated on STN: 19911013 Entered Medline: 19910924

AB A new solid-phase extraction method using octyl-silica columns to extract vasopressin-like immunoreactivity from plasma has been developed. The extraction was followed by a radioimmunoassay on the vacuum-dried extracts, which were reconstituted in assay buffer. The total recovery of synthetic vasopressin was ca. 100%. Based

on co-elution with synthetic vasopressin after **separation** by reversed-**phase** high-performance **liquid** chromatography of plasma extracts from normal Wistar and Brattleboro rats, and the

reactivity of the antiserum used in the radioimmunoassay
system, the extracted material was found to be indistinguishable from
authentic vasopressin. Unknown experimental samples were interpolated on

standard curve established in "zero" plasma (plasma derived from rats subjected to waterload) spiked with known amounts of synthetic vasopressin, and not on a standard curve established in **assay** buffer. The limit of detection was 1 fmol of vasopressin equivalent per millilitre. The intra- and inter-**assay** coefficients of variance were 10-16% and 16%, respectively. The procedure reliably showed that osmotic challenge and 24-h dehydration increased, whereas ethanol ingestion decreased vasopressin-like immunoreactivity plasma levels in

the

rat, compared with normally hydrated controls.

TI **Solid-phase** extraction of plasma vasopressin: evaluation, validation and application.

AB A new solid-phase extraction method using octyl-silica columns to extract vasopressin-like immunoreactivity from plasma has been developed. The extraction was followed by a radioimmunoassay on the vacuum-dried extracts, which were reconstituted in assay buffer. The total recovery of synthetic vasopressin was ca. 100%. Based

co-elution with synthetic vasopressin after **separation** by reversed-**phase** high-performance **liquid** chromatography of plasma extracts from normal Wistar and Brattleboro rats, and the

reactivity of the antiserum used in the radioimmunoassay
system, the extracted material was found to be indistinguishable from
authentic vasopressin. Unknown experimental samples were interpolated on

on

standard. . . from rats subjected to waterload) spiked with known amounts of synthetic vasopressin, and not on a standard curve established in assay buffer. The limit of detection was 1 fmol of

vasopressin equivalent per millilitre. The intra- and inter-assay coefficients of variance were 10-16% and 16%, respectively. The procedure reliably showed that osmotic challenge and 24-h dehydration increased, whereas. . .

Check Tags: Animal; Male CT

\*Chromatography, High Pressure Liquid: MT, methods

Radioimmunoassay

Rats

Rats, Inbred BB Rats, Inbred Strains

\*Vasopressins: BL, blood

L11 ANSWER 1 OF 40 USPATFULL

AN 2001:79295 USPATFULL

TI Energy transfer hybridization assay composition

IN Rabbani, Elazar, New York, NY, United States Hurley, Ian, Staten Island, NY, United States

PA Enzo Diagnostics, Inc., Farmingdale, NY, United States (U.S.

corporation)

PI US 6239271 B1 20010529

AI US 1999-386695 19990831 (9)

RLI Continuation of Ser. No. US 1995-486053, filed on 7 Jun 1995, now patented, Pat. No. US 5998135, issued on 7 Dec 1999 Continuation of

Ser.

No. US 1994-194215, filed on 9 Feb 1994, now abandoned Continuation of

Ser. No. US 1989-314995, filed on 24 Feb 1989, now abandoned

DT Utility

LN.CNT 740

INCL INCLM: 536/024.300 NCL NCLM: 536/024.300

IC [7]

ICM: C07H021-04

EXF 536/24.3; 435/6; 435/810; 436/94 CAS INDEXING IS AVAILABLE FOR THIS PATENT.

## => d 111 ibib ab 1-40

L11 ANSWER 1 OF 40 USPATFULL

ACCESSION NUMBER: 2001:79295 USPATFULL

TITLE:
INVENTOR(S):

Energy transfer hybridization assay composition Rabbani, Elazar, New York, NY, United States Hurley, Ian, Staten Island, NY, United States

PATENT ASSIGNEE(S):

Enzo Diagnostics, Inc., Farmingdale, NY, United States

(U.S. corporation)

APPLICATION INFO.:

US 1999-386695 19990831 (9)

RELATED APPLN. INFO.:

FO.: Continuation of Ser. No. US 1995-486053, filed on 7

1995, now patented, Pat. No. US 5998135, issued on 7 Dec 1999 Continuation of Ser. No. US 1994-194215,

filed

on 9 Feb 1994, now abandoned Continuation of Ser. No. US 1989-314995, filed on 24 Feb 1989, now abandoned

DOCUMENT TYPE: Utility

PRIMARY EXAMINER: Horlick, Kenneth R.

LEGAL REPRESENTATIVE: Fedus, Esq., Ronald C., Rogers, Esq., James L.

NUMBER OF CLAIMS: 22 EXEMPLARY CLAIM: 1 LINE COUNT: 740

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

Disclosed is a nucleic acid hybridization assay composition for detecting the presence or absence of a target oligo- or polynucleotide in a sample. The composition comprises: a solid matrix having at least one surface which is substituted with a first intercalator capable of binding dsDNA, dsRNA, or DNA-RNA hybrids; a second intercalator, which may or may not comprise at least one fluorophore, said intercalator or

said fluorophore, each acting as either an energy donor or an energy acceptor; and an oligo- or polynucleotide probe which is specifically hybridizable with the target oligo- or polynucleotide and has directly or indirectly bound thereto, at least one lanthanide metal chelate or

at

least one fluorophore, each acting as either an energy donor or an energy acceptor. Also disclosed are a method and kit for its use.

L11 ANSWER 2 OF 40 USPATFULL

ACCESSION NUMBER: 2001:78896 USPATFULL

TITLE: High throughput assay system

INVENTOR(S): Kris, Richard M, Tucson, AZ, United States Felder, Stephen, Tucson, AZ, United States

PATENT ASSIGNEE(S): High Throughput Genomics, Inc., Tucson, AZ, United

States (U.S. corporation)

NUMBER KIND DATE

PATENT INFORMATION: US 6238869
APPLICATION INFO.: US 1999-33732

US 6238869 B1 20010529 US 1999-337325 19990621 (9)

RELATED APPLN. INFO.: Continuation-in-part of Ser. No. US 1998-218166, filed

on 22 Dec 1998, now abandoned

NUMBER DATE

PRIORITY INFORMATION: US 1997-68291 19971219 (60)

DOCUMENT TYPE: Utility

PRIMARY EXAMINER: Brusca, John S. ASSISTANT EXAMINER: Kim, Young

LEGAL REPRESENTATIVE: Millen, White, Zelano & Branigan

NUMBER OF CLAIMS: 32 EXEMPLARY CLAIM: 1

NUMBER OF DRAWINGS: 28 Drawing Figure(s); 22 Drawing Page(s)

LINE COUNT: 3409

AB The present invention relates to compositions, apparatus and methods useful for concurrently performing multiple, high throughput, biological

or chemical assays, using repeated arrays of probes. A combination of the invention comprises a surface, which comprises a plurality of test regions, at least two of which, and in a preferred embodiment, at least twenty of which, are substantially identical, wherein each of the test regions comprises an array of generic anchor molecules. The anchors are associated with bifunctional linker molecules, each containing a

portion

which is specific for at least one of the anchors and a portion which is

a probe specific for a target of interest. The resulting array of probes

is used to analyze the presence or test the activity of one or more target molecules which specifically interact with the probes. In one embodiment of the invention, the test regions (which can be wells) are further subdivided into smaller subregions (indentations, or dimples).

L11 ANSWER 3 OF 40 USPATFULL

ACCESSION NUMBER: 2001:75129 USPATFULL

TITLE: Process for detecting nucleic acids by mass

determination

INVENTOR(S): Bergmann, Frank, Iffeldorf, Germany, Federal Republic

οf

Herrmann, Rupert, Weilheim, Germany, Federal Republic

ΟÍ

Kobold, Uwe, Wielenbach, Germany, Federal Republic of

PATENT ASSIGNEE(S): Dako A/S, Glostrup, Denmark (non-U.S. corporation)

NUMBER KIND DATE

US 6235476 WO 9807885 PATENT INFORMATION: 20010522

19980226 19990317 (9)

US 1999-242536 APPLICATION INFO.: WO 1997-EP4494 19970818

> 19990317 PCT 371 date 19990317 PCT 102(e) date

NUMBER DATE -----PRIORITY INFORMATION: DE 1996-19633436 19960820

DOCUMENT TYPE: Utility
PRIMARY EXAMINER: Horlick, Kenneth R.

LEGAL REPRESENTATIVE: Arent Fox Kintner Plotkin Kahn PLLC.

NUMBER OF CLAIMS: 17 EXEMPLARY CLAIM:

NUMBER OF DRAWINGS: 3 Drawing Figure(s); 3 Drawing Page(s)

LINE COUNT: 619

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

A method for detecting nucleic acids by binding a probe P to a partial sequence S contained in the nucleic acid to produce a binding product

B1, degrading the nucleic acid to produce a binding product B2

containing a partial nucleic acid F of a defined length and detecting the binding product B2 or the partial nucleic acid F a part based on

its

mass is particularly suitable for the parallel detection of nucleic

acid

of different sequences.

L11 ANSWER 4 OF 40 USPATFULL

2001:71297 USPATFULL ACCESSION NUMBER:

AC methods for the detection of nucleic acids TITLE: Kayyem, Jon Faiz, Pasadena, CA, United States INVENTOR(S):

O'Connor, Stephen D., Pasadena, CA, United States

Clinical Micro Sensors, Inc., Pasadena, CA, United PATENT ASSIGNEE(S):

States (U.S. corporation)

NUMBER KIND DATE \_\_\_\_\_

US 6232062 B1 20010515 US 1997-911589 19970814 (8) PATENT INFORMATION:
APPLICATION INFO.: APPLICATION INFO.:

Continuation of Ser. No. US 1997-873597, filed on 12 RELATED APPLN. INFO.:

Jun 1997

NUMBER DATE US 1997-40155 19970307 (60) PRIORITY INFORMATION:

DOCUMENT TYPE: Utility

PRIMARY EXAMINER: Whisenant, Ethan

LEGAL REPRESENTATIVE: Flehr Hohbach Test Albritton & Herbert LLP, Trecartin,

Richard F., Silva, Robin M.

30 NUMBER OF CLAIMS: 1 EXEMPLARY CLAIM:

NUMBER OF DRAWINGS: 56 Drawing Figure(s); 39 Drawing Page(s)

4220 LINE COUNT:

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

The invention relates to nucleic acids covalently coupled to electrodes via conductive oligomers. More particularly, the invention is directed

to the site-selective modification of nucleic acids with electron

transfer moieties and electrodes to produce a new class of

biomaterials,

and to methods of making and using them.

L11 ANSWER 5 OF 40 USPATFULL

ACCESSION NUMBER: 2001:51852 USPATFULL

Method of conducting an assay of a sample containing TITLE:

an

analyte of interest

Lakowicz, Joseph R., 10037 Fox Den Rd., Ellicott City, INVENTOR(S):

MD, United States 21042

Castellano, Felix, Columbia, MD, United States Murtaza, Zakir, Baltimore, MD, United States

Lakowicz, Joseph R., Ellicott City, MD, United States PATENT ASSIGNEE(S):

(U.S. individual)

NUMBER KIND DATE \_\_\_\_\_

PATENT INFORMATION: US 6214628 B1 20010410 APPLICATION INFO.: US 1998-7167 19980114 19980114 (9) APPLICATION INFO.:

Utility DOCUMENT TYPE: PRIMARY EXAMINER: Le, Long V.
ASSISTANT EXAMINER: Cook, Lisa J.

LEGAL REPRESENTATIVE: Rothwell, Figg, Ernst & Manbeck, p.c.

NUMBER OF CLAIMS: 15 1 EXEMPLARY CLAIM:

NUMBER OF DRAWINGS: 39 Drawing Figure(s); 29 Drawing Page(s)

1382 LINE COUNT:

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

In accordance with the present invention, a method of conducting an assay of a sample containing an analyte of interest includes the step of

forming a mixture so as to bring a metal-ligand complex into interactive

proximity with the sample containing the analyte of interest. The mixture is irradiated with electromagnetic light energy so as to cause emission of light indicative of the analyte of interest. The emitted light is measured, and the measurement of the emitted light is utilized to measure the analyte of interest. The metal-ligand complex can be [Re(bcp)(CO).sub.3 (4-COOHPy)].sup.+, [Os(phen).sub.2 (aphen)].sup.2+, [Os(tpy)(triphos)].sup.2+, [Os(tppz).sub.2].sup.2+, and

[Os(ttpy).sub.2 1.sup.2+, or the like. Also, the present invention is directed to a metal-ligand complex of the formula [Re(bcp)(CO).sub.3 (4-COOHPy)].sup.+.

L11 ANSWER 6 OF 40 USPATFULL

2001:44388 USPATFULL ACCESSION NUMBER:

Fluorescent dyes (AIDA) for solid phase and solution TITLE:

phase screening

Auer, Manfred, Moedling, Austria INVENTOR (S):

Gstach, Hubert, Vienna, Austria

Novartis AG, Basel, Switzerland (non-U.S. corporation) PATENT ASSIGNEE(S):

NUMBER KIND DATE

PATENT INFORMATION: US 6207831 B1 20010327
APPLICATION INFO.: US 1998-217795 19981221 (9)

DOCUMENT TYPE: Utility

PRIMARY EXAMINER: Lambkin, Deborah C. ASSISTANT EXAMINER: Wright, Sonya LEGAL REPRESENTATIVE: Lopez, Gabriel

NUMBER OF CLAIMS: EXEMPLARY CLAIM:

4 Drawing Figure(s); 4 Drawing Page(s) NUMBER OF DRAWINGS:

1831 LINE COUNT:

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

The invention relates to new fluorescent dyes of formula (I) ##STR1## AB

which can be used in high throughput screening both, on the solid phase as well as in homogeneous solution.

L11 ANSWER 7 OF 40 USPATFULL

ACCESSION NUMBER: 2001:29375 USPATFULL

Methods for monitoring the status of assays and TITLE:

immunoassays

Buechler, Kenneth F., San Diego, CA, United States INVENTOR(S):

Anderberg, Joseph M, Encinitas, CA, United States McPherson, Paul H., Encinitas, CA, United States

PATENT ASSIGNEE(S):

Biosite Diagnostics, Inc., San Diego, CA, United

States

(U.S. corporation)

NUMBER KIND DATE \_\_\_\_\_

US 6194222 B1 20010227 US 1998-3065 19980105 PATENT INFORMATION: APPLICATION INFO.: 19980105 (9)

DOCUMENT TYPE: Utility
PRIMARY EXAMINER: Le, Long V.
ASSISTANT EXAMINER: Cook, Lisa V.
NUMBER OF CLAIMS: 28
EXEMPLARY CLAIM: 1

EXEMPLARY CLAIM: 1
NUMBER OF DRAWINGS: 18 Drawing Figure(s); 18 Drawing Page(s)
LINE COUNT: 3661

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

The invention relates in part to the use of independent assay controls (IACs) for the optical communication between an assay device and an

instrument in monitoring and performing assays, preferably

immunoassays.

L11 ANSWER 8 OF 40 USPATFULL

ACCESSION NUMBER: 2000:157233 USPATFULL

High throughput screening assay systems in microscale TITLE:

fluidic devices

Parce, John Wallace, Palo Alto, CA, United States INVENTOR (S):

Kopf-Sill, Anne R., Portola Valley, CA, United States

Bousse, Luc J., Menlo Park, CA, United States

Caliper Technologies Corp., Mountain View, CA, United PATENT ASSIGNEE(S):

States (U.S. corporation)

NUMBER KIND DATE \_\_\_\_\_\_ US 6150180 20001121 US 1999-360782 19990726 (9)

PATENT INFORMATION:
APPLICATION INFO.: Continuation of Ser. No. US 1996-671987, filed on 28 RELATED APPLN. INFO.:

Jun 1996, now patented, Pat. No. US 5942443

NUMBER DATE \_\_\_\_\_

US 1996-15498 19960416 (60) PRIORITY INFORMATION:

Utility

DOCUMENT TYPE: PRIMARY EXAMINER: PRIMARY EXAMINER: Chin, Christopher I ASSISTANT EXAMINER: Pham, Minh-Quan K. Chin, Christopher L.

LEGAL REPRESENTATIVE: Murphy, Matthew B., Shaver, Gulshan H.

14 NUMBER OF CLAIMS: 1 EXEMPLARY CLAIM:

17 Drawing Figure(s); 14 Drawing Page(s) 1480 NUMBER OF DRAWINGS:

LINE COUNT:

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

The present invention provides novel microfluidic devices and methods that are useful for performing high-throughput screening assays. In particular, the devices and methods of the invention are useful in screening large numbers of different compounds for their effects on a variety of chemical, and preferably, biochemical systems.

L11 ANSWER 9 OF 40 USPATFULL

ACCESSION NUMBER: 2000:131592 USPATFULL

Detection of nucleic acids and nucleic acid units TITLE:

Graham, Duncan, Edinburgh, United Kingdom INVENTOR(S):

Linacre, Adrian Matthew Thornton, Glasgow, United

Munro, Callum Hugh, Pittsburgh, PA, United States Smith, William Ewan, Glasgow, United Kingdom Watson, Nigel Dean, Ayrshire, United Kingdom White, Peter Cyril, Drymen, United Kingdom

PATENT ASSIGNEE(S):

University of Strathclyde, Glasgow, United Kingdom

(non-U.S. corporation)

	NUMBER	KIND	DATE	
•				
PATENT INFORMATION:	US 6127120		20001003	
	WO 9705280		19970213	
APPLICATION INFO.:	US 1998-983486		19980421	(8)
	WO 1996-GB1830		19960725	

19980421 PCT 371 date 19980421 PCT 102(e) date

NUMBER DATE \_\_\_\_\_\_

PRIORITY INFORMATION:

GB 1995-17955

19950725

Utility DOCUMENT TYPE: PRIMARY EXAMINER:

Riley, Jezia

Dann, Dorfman, Herrell and Skillman

LEGAL REPRESENTATIVE: 47 NUMBER OF CLAIMS: EXEMPLARY CLAIM:

1

NUMBER OF DRAWINGS: LINE COUNT:

22 Drawing Figure(s); 22 Drawing Page(s)

2282

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

The invention relates to the detection of target nucleic acids or nucleic acid units in a sample, by obtaining a SER(R)S spectrum for a SER(R)S-active complex containing, or derived directly from, the

target.

The complex includes at least a SER(R)S-active label, and optionally a target binding species containing a nucleic acid or nucleic acid unit. In this detection method, the concentration of the target present in

the

SER(R)S-active complex, or of the nucleic acid or unit contained in the target binding species in the SER(R)S-active complex, is no higher than 10.sup.-10 moles per liter. Additionally or alternatively, one or more of the following features may be used with the method: i) the introduction of a polyamine; ii) modification of the target, and/or of the nucleic acid or nucleic acid unit contained in the target binding species, in a manner that promotes or facilitates its chemi-sorption onto a SER(R)S-active surface; iii) inclusion of a chemi-sorptive functional group in the SER(R)S-active label. The invention also provides SER(R)S-active complexes for use in such a method, a kit for use in carrying out the method or preparing the complexes and a method for sequencing a nucleic acid which comprises the use of the detection method to detect at least one target nucleotide or sequence of nucleotides within the acid.

L11 ANSWER 10 OF 40 USPATFULL

2000:124833 USPATFULL ACCESSION NUMBER:

Methods and devices for conducting specific binding TITLE:

INVENTOR(S):

Hargreaves, William R., Bellevue, WA, United States

PATENT ASSIGNEE(S):

Roche Diagnostics Corporation, Indianapolis, IN,

KIND DATE

United

States (U.S. corporation)

	.,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,		
PATENT INFORMATION:	US 6121055	20000919	
APPLICATION INFO.:	US 1995-430265	19950428 (8)	
DELYGED ADDIN INFO .	Continuation of Ser	No. US 1994-200944.	

NIMBER

RELATED APPLN. INFO.: Feb 1994, now abandoned which is a continuation of

Ser.

No. US 1991-687850, filed on 19 Apr 1991 which is a division of Ser. No. US 1987-127944, filed on 1 Dec 1987, now abandoned which is a continuation-in-part of Ser. No. US 1995-768108, filed on 21 Aug 1995, now

abandoned

DOCUMENT TYPE:

Utility

PRIMARY EXAMINER: LEGAL REPRESENTATIVE:

Chin, Christopher L. Seed IP Law Group

NUMBER OF CLAIMS: EXEMPLARY CLAIM:

NUMBER OF DRAWINGS:

6 Drawing Figure(s); 3 Drawing Page(s)

LINE COUNT:

2789

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

The invention provides methods for separating bound label from unbound label within an assay mixture, for predispensing assay reactants in self-contained assay vessels, as well as for detecting the presence and/or amount of an analyte within a fluid sample. In addition, a reusable detection vessel for use therein and with specific binding assays in general is disclosed. In the methods, generally an analyte within a sample is detected or measured by forming an assay mixture

containing sample, analyte binding components and label, placing the assay mixture in contact with an immisable primary layer, subjecting

the

assay mixture to conditions that separate the analyte bound with binding

components and label from unbound binding components and label, and subsequently detecting bound label.

L11 ANSWER 11 OF 40 USPATFULL

ACCESSION NUMBER:

2000:121769 USPATFULL

TITLE:

Method for enhancing fluorescence

INVENTOR(S):

Zanzucchi, Peter John, Lawrenceville, NJ, United

States

PATENT ASSIGNEE(S):

Sarnoff Corporation, Princeton, NJ, United States

(U.S.

corporation)

NUMBER KIND DATE \_\_\_\_\_\_

PATENT INFORMATION:

US 6118126 20000912

APPLICATION INFO.:

US 1998-187355 19981106 (9)

RELATED APPLN. INFO.:

Continuation-in-part of Ser. No. US 1997-961860, filed

on 31 Oct 1997

DOCUMENT TYPE:

Utility

PRIMARY EXAMINER: ASSISTANT EXAMINER:

Minnifield, Nita Baskar, Padma

LEGAL REPRESENTATIVE: NUMBER OF CLAIMS:

Burke, William J.

EXEMPLARY CLAIM:

13

NUMBER OF DRAWINGS:

5 Drawing Figure(s); 2 Drawing Page(s)

LINE COUNT:

1538

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

The present invention relates to a method for the enhancement of AB fluorescence wherein a fluorophor is connected to a textured material. The method can be used in any forensic or medical diagnostic assay, particularly where the absence or presence of a molecule having a concentration of less than about 1 .mu.g/ml is desirably determined.

L11 ANSWER 12 OF 40 USPATFULL

ACCESSION NUMBER:

2000:92102 USPATFULL

TITLE:

Method for the synthesis of pyrrole and imidazole

carboxamides on a solid support

INVENTOR(S):

Dervan, Peter B., San Marino, CA, United States

Baird, Eldon, Pasadena, CA, United States

PATENT ASSIGNEE(S):

California Institute of Technology, Pasadena, CA,

United States (U.S. corporation)

NUMBER KIND DATE

PATENT INFORMATION: US 6090947 20000718 APPLICATION INFO.: US 1996-607078 19960226 (8)

DOCUMENT TYPE: Utility

PRIMARY EXAMINER: Higel, Floyd D.
LEGAL REPRESENTATIVE: Lyon & Lyon LLP

NUMBER OF CLAIMS: 49 EXEMPLARY CLAIM: 1

NUMBER OF DRAWINGS: 50 Drawing Figure(s); 48 Drawing Page(s)

LINE COUNT: 4496

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

The present invention describes a novel method for the solid phase synthesis of polyamides containing imidazole and pyrrole carboxamides. The polyamides are prepared on a solid support from aromatic carboxylic acids and aromatic amines with high stepwise coupling yields (>99%), providing milligram quantities of highly pure polyamides. The present invention also describes the synthesis of analogs of the natural products Netropsin and Distamycin A, two antiviral antibiotics. The present invention also describes a novel method for the solid phase synthesis of imidazole and pyrrole carboxamide

polyamide-oligonucleotide

conjugates. This methodology will greatly increase both the complexity and quantity of minor-groove binding polyamides and minor-groove binding

polyamide-oligonucleotide conjugates which can be synthesized and tested.

L11 ANSWER 13 OF 40 USPATFULL

ACCESSION NUMBER: 2000:92088 USPATFULL

TITLE: Methods of attaching conductive oligomers to

electrodes

INVENTOR(S): Kayyem, Jon Faiz, Pasadena, CA, United States

O'Connor, Stephen D., Pasadena, CA, United States

Gozin, Michael, Beer Sheva, Israel

Yu, Changjun, Pasadena, CA, United States Meade, Thomas J., Altadena, CA, United States

PATENT ASSIGNEE(S): Clinical Micro Sensors, Inc., Pasadena, CA, United

States (U.S. corporation)

RELATED APPLN. INFO.: Continuation of Ser. No. US 1997-873978, filed on 12

Jun 1997 which is a continuation of Ser. No. US

1996-743798, filed on 5 Nov 1996

DOCUMENT TYPE: Utility

PRIMARY EXAMINER: Marschel, Ardin H.

LEGAL REPRESENTATIVE: Trecartin, Esq., Richard F., Silva, Esq., Robin

M.Flehr

Hohbach Test Albritton & Herbert LLP

NUMBER OF CLAIMS: 11 EXEMPLARY CLAIM: 1

NUMBER OF DRAWINGS: 44 Drawing Figure(s); 39 Drawing Page(s)

LINE COUNT: 4152

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB The invention relates to nucleic acids covalently coupled to electrodes via conductive oligomers. More particularly, the invention is directed to the site-selective modification of nucleic acids with electron

transfer moieties and electrodes to produce a new class of

biomaterials,

and to methods of making and using them.

L11 ANSWER 14 OF 40 USPATFULL

ACCESSION NUMBER: 2000:61453 USPATFULL

TITLE: Sensors for sugars and other metal binding analytes INVENTOR(S):

Arnold, Frances H., Pasadena, CA, United States Guan, Zhibin, Hockessin, DE, United States

Chen, Chao-Tsen, New York, NY, United States Chen, Guohua, Pasadena, CA, United States

California Institute of Technology, Pasadena, CA, PATENT ASSIGNEE(S):

United States (U.S. corporation)

NUMBER KIND DATE \_\_\_\_\_\_ US 6063637 20000516 PATENT INFORMATION: WO 9733177 19970912 US 1997-875047 APPLICATION INFO.: 19970707 (8) WO 1997-US3654 19970303

19970707 PCT 371 date

19970707 PCT 102(e) date RELATED APPLN. INFO.: Continuation-in-part of Ser. No. US 1995-571440, filed

on 13 Dec 1995, now abandoned

NUMBER DATE

PRIORITY INFORMATION: US 1996-12756 19960304 (60)

Utility

DOCUMENT TYPE:
PRIMARY EXAMINER: Soderquist, Arlen LEGAL REPRESENTATIVE: Darby & Darby

NUMBER OF CLAIMS: 32 EXEMPLARY CLAIM: 17

NUMBER OF DRAWINGS: 38 Drawing Figure(s); 24 Drawing Page(s) LINE COUNT: 2889

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

Sensors (20, 50, 70) for use in detecting the presence of sugars and other analytes (target molecules). The sensor is composed of a metal complex that binds to the target molecule and releases a proton or includes an exchangable ligand which is exchanged for the target molecule during the binding interaction between the metal complex and the target molecule. The result of the binding interaction is the release of a proton, hydroxide ion or ligand species generated during the ligand exchange. Measurement of the release of proton, hydroxide

ion

or other ligand species from the sensor (20, 50, 70) provides an indirect indication of target molecule concentration. The metal complexes may be attached to support structures (10, 12) to provide

both

anchoring and positioning of the metal ions to increase selectivity of sugar/metal complex interactions. Detection systems in which pH is used as an indication of proton or hydroxide release are disclosed, as are detection systems in which Cl.sup.- release is used. Methods for monitoring the concentrations of sugars and related molecules using the metal based sensors (20, 50, 70) are also disclosed.

L11 ANSWER 15 OF 40 USPATFULL

2000:61390 USPATFULL ACCESSION NUMBER:

TITLE: Cycling probe technology using electron transfer

INVENTOR (S): Kayyem, Jon Faiz, Pasadena, CA, United States

Clinical Micro Sensors, Inc., Pasadena, CA, United PATENT ASSIGNEE(S):

States (U.S. corporation)

NUMBER KIND DATE US 6063573 20000516 US 1998-14304 19980127 (9) PATENT INFORMATION: US 6063573
APPLICATION INFO.: US 1998-14304

DOCUMENT TYPE: Utility

PRIMARY EXAMINER: Marschel, Ardin H.

LEGAL REPRESENTATIVE: Flehr Hohbach Test Albritton & Herbert LLP, Silva,

Robin M., Trecartin, Richard F.

NUMBER OF CLAIMS: EXEMPLARY CLAIM:

53 Drawing Figure(s); 48 Drawing Page(s) NUMBER OF DRAWINGS:

LINE COUNT: 4975

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

The invention relates to novel methods and compositions useful in AB

Cycling Probe Technology (CPT) using electron transfer to detect target

nucleic acid sequences.

L11 ANSWER 16 OF 40 USPATFULL

2000:47094 USPATFULL ACCESSION NUMBER:

Optical chemical sensor based on multilayer TITLE:

self-assembled thin film sensors for aquaculture

process control

Luo, Shufang, Blacksburg, VA, United States INVENTOR(S):

Lo, K. Peter, Blacksburg, VA, United States

Groger, Howard P., Gainesville, FL, United States Churchill, Russell J., Radford, VA, United States American Research Corporation of Virginia, Radford,

PATENT ASSIGNEE(S):

VA.

United States (U.S. corporation)

NUMBER KIND DATE US 6051437 20000418 PATENT INFORMATION: PATENT INFORMATION:
APPLICATION INFO.: US 1998-71775 19980504 (9) Utility

DOCUMENT TYPE:

PRIMARY EXAMINER: Snay, Jeffrey

LEGAL REPRESENTATIVE: Wray, James Creighton, Narasimhan, Meera P.

NUMBER OF CLAIMS: EXEMPLARY CLAIM: 1

NUMBER OF DRAWINGS: 46 Drawing Figure(s); 45 Drawing Page(s)

LINE COUNT: 1990

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

Optical chemical probes have layers of anionic and cationic polyelectrolytes and one or more dyes incorporated into these layers. The probes are placed into the medium and the dye or dyes react in the presence of the corresponding chemical. Color changes may be observed manually or by a photo detector. A light source may be employed to increase the optical signal received from the probe. Further, a

waveguide may be used to trap multiple optical signals. The invention

is

used for chemical analysis.

L11 ANSWER 17 OF 40 USPATFULL

2000:40901 USPATFULL ACCESSION NUMBER:

TITLE:

High throughput screening assay systems in microscale

fluidic devices

Parce, J. Wallace, Palo Alto, CA, United States INVENTOR (S):

Kopf-Sill, Anne R., Portola Valley, CA, United States

Bousse, Luc J., Menlo Park, CA, United States

PATENT ASSIGNEE(S):

Caliper Technologies Corporation, Palo Alto, CA,

United

States (U.S. corporation)

NUMBER KIND DATE \_\_\_\_\_\_ US 6046056 20000404 US 1996-761575 19961206 (8) PATENT INFORMATION:

APPLICATION INFO.:

Continuation-in-part of Ser. No. US 1996-671987, filed RELATED APPLN. INFO.:

on 28 Jun 1996

NUMBER DATE \_\_\_\_\_

US 1996-15498 19960416 (60) PRIORITY INFORMATION:

Utility DOCUMENT TYPE:

DOCUMENT TYPE: Utility
PRIMARY EXAMINER: Chin, Christopher L.

LEGAL REPRESENTATIVE: Townsend and Townsend and Crew, LLP, Murphy, Matthew

B., Quine, Jonathan Alan

NUMBER OF CLAIMS: 38 EXEMPLARY CLAIM: 1

NUMBER OF DRAWINGS: 20 Drawing Figure(s); 17 Drawing Page(s)

LINE COUNT: 1669

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

The present invention provides novel microfluidic devices and methods that are useful for performing high-throughput screening assays. In particular, the devices and methods of the invention are useful in screening large numbers of different compounds for their effects on a

variety of chemical, and preferably, biochemical systems.

L11 ANSWER 18 OF 40 USPATFULL

ACCESSION NUMBER: 2000:34437 USPATFULL

TITLE: Raman spectroscopic method for determining the ligand

binding capacity of biologicals

Grow, Ann E., 5882 Highplace Dr., San Diego, CA, INVENTOR(S):

United

States 92120

NUMBER KIND DATE

US 6040191 20000321 US 1998-177548 19981022 (9) PATENT INFORMATION: APPLICATION INFO.:

Division of Ser. No. US 1997-864015, filed on 27 May RELATED APPLN. INFO.:

1997, now patented, Pat. No. US 5866430

NUMBER DATE \_\_\_\_\_

US 1996-19742 19960613 (60) PRIORITY INFORMATION:

DOCUMENT TYPE: Utility
PRIMARY EXAMINER: Weber, Jon P. LEGAL REPRESENTATIVE: Beehler & Pavitt

NUMBER OF CLAIMS: 11 1 EXEMPLARY CLAIM:

NUMBER OF DRAWINGS: 20 Drawing Figure(s); 8 Drawing Page(s)

LINE COUNT: 3869

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB A nondestructive process for determining the reactive capacity of a test

biological by Raman scattering. The test biological may be any one of enzymes, enzyme cofactors, coenzymes, antibodies, antibody fragments, hemeproteins, peptides, synthetic peptides, toxins, toxoids, glycosphingolipids, lectins, lipids, lipid complexes, phospholipids, carbohydrates, saccharides, gangliosides, nucleic acids, fragments of nucleic acids, pathogen adhesion factors, receptors, receptor subunits, membranes, organelles, cells, tissues and complexes containing membranes, organelles, cells and tissues, or a bioconcentrator. The

test

biological is irradiated with a light source to produce a Raman scattering spectrum of the irradiated biological. The Raman scattering spectrum is collected and processed to determine the ability of the

test

biological to react with ligands. The analyzing step includes comparing the Raman scattering spectrum of the test biological against that of a biological standard of the same biological which has been altered to vary the capability to react with ligands thereby determining the capacity of the test biological to react with ligands.

L11 ANSWER 19 OF 40 USPATFULL

2000:2826 USPATFULL ACCESSION NUMBER:

Detection of biological molecules using chemical TITLE:

amplification and optical sensors

Van Antwerp, William Peter, Valencia, CA, United INVENTOR(S):

States

Mastrototaro, John Joseph, Los Angeles, CA, United

States

Minimed Inc., Sylmar, CA, United States (U.S. PATENT ASSIGNEE(S):

corporation)

NUMBER KIND DATE \_\_\_\_\_\_

PATENT INFORMATION: US 6011984 20000104 US 1996-752945 19961121 (8) APPLICATION INFO.:

> NUMBER DATE -----

US 1996-7515 19960926 (60) PRIORITY INFORMATION:

DOCUMENT TYPE:

PRIMARY EXAMINER:

LEGAL REPRESENTATIVE:

Utility

Winakur, Eric F.

Townsend and Townsend and Crew LLP

NUMBER OF CLAIMS: 14 1 EXEMPLARY CLAIM:

NUMBER OF DRAWINGS: 16 Drawing Figure(s); 14 Drawing Page(s)

LINE COUNT: 1474

Methods are provided for the determination of the concentration of AΒ biological levels of polyhydroxylated compounds, particularly glucose. The methods utilize an amplification system that is an analyte

transducer immobilized in a polymeric matrix, where the system is implantable and biocompatible. Upon interrogation by an optical system,

the amplification system produces a signal capable of detection

to the skin of the patient. Quantitation of the analyte of interest is achieved by measurement of the emitted signal.

L11 ANSWER 20 OF 40 USPATFULL

ACCESSION NUMBER: 1999:164806 USPATFULL

Detection of biological molecules using boronate-based TITLE:

> chemical amplification and optical sensors Van Antwerp, William Peter, Valencia, CA, United

INVENTOR(S):

States

Mastrototaro, John Joseph, Los Angeles, CA, United

States

Lane, Stephen M., Oakland, CA, United States Satcher, Jr., Joe H., Modesto, CA, United States Darrow, Christopher B., Pleasanton, CA, United States

Peyser, Thomas A., Menlo Park, CA, United States Harder, Jennifer, Livermore, CA, United States

The Regents of the University of California, Oakland, PATENT ASSIGNEE(S):

CA, United States (U.S. corporation)

Minimed Inc., Sylmar, CA, United States (U.S.

corporation)

NUMBER KIND DATE \_\_\_\_\_ PATENT INFORMATION: US 6002954 US 1996-749366 19991214

19961121 (8) APPLICATION INFO.:

> NUMBER DATE \_\_\_\_\_

US 1995-7515 19951122 (60) PRIORITY INFORMATION:

Utility

DOCUMENT TYPE: PRIMARY EXAMINER: Winakur, Eric F.

LEGAL REPRESENTATIVE: Townsend and Townsend and Crew LLP NUMBER OF CLAIMS: 35

NUMBER OF CLAIMS: EXEMPLARY CLAIM:

NUMBER OF DRAWINGS: 18 Drawing Figure(s); 14 Drawing Page(s)

1565 LINE COUNT:

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

Methods are provided for the determination of the concentration of biological levels of polyhydroxylated compounds, particularly glucose. The methods utilize an amplification system that is an analyte transducer immobilized in a polymeric matrix, where the system is implantable and biocompatible. Upon interrogation by an optical system, the amplification system produces a signal capable of detection

to the skin of the patient. Quantitation of the analyte of interest is achieved by measurement of the emitted signal.

L11 ANSWER 21 OF 40 USPATFULL

ACCESSION NUMBER: 1999:159751 USPATFULL

TITLE: Energy transfer hybridization assay using

intercalators

external

and lanthanide metals

INVENTOR(S): Rabbani, Elazar, New York, NY, United States
Hurley, Ian, Staten Island, NY, United States

PATENT ASSIGNEE(S): Enzo Diagnostics, Inc., Farmingdale, NY, United States

(U.S. corporation)

NUMBER KIND DATE

PATENT INFORMATION: US 5998135 19991207 APPLICATION INFO.: US 1995-486053 19950607 (8)

RELATED APPLN. INFO.: Continuation of Ser. No. US 1994-194215, filed on 9

Feb

1994, now abandoned which is a continuation of Ser.

No.

US 1989-314995, filed on 24 Feb 1989, now abandoned

DOCUMENT TYPE: Utility

PRIMARY EXAMINER: Horlick, Kenneth R. LEGAL REPRESENTATIVE: Fedus, Esq., Ronald C.

NUMBER OF CLAIMS: 53 EXEMPLARY CLAIM: 26 LINE COUNT: 876

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

Disclosed is a nucleic acid hybridization assay composition for detecting the presence of absence of a target oligo- or polynucleotide in a sample. The composition comprises: a solid matrix having at least one surface which is substituted with a first intercalator capable of binding dsDNA dsRNA, or DAN-RNA hybrids; a second intercalator, which may or may not comprise at least one fluorophore, said intercalator or said fluorophore each acting as either an energy donor or an energy acceptor; and an oligo- or polynucleotide probe which is specifically hybridizable with the target oligo- or polynucleotide and has directly or indirectly bound thereto, at least one lanthanide metal chelate or

at

least one fluorophore, each acting as either an energy donor or an energy acceptor. Also disclosed are a method and kit for its use.

L11 ANSWER 22 OF 40 USPATFULL

ACCESSION NUMBER: 1999:137045 USPATFULL

TITLE: Immunoassays in capillary tubes

INVENTOR(S): Kumar, Amit, Milpitas, CA, United States

Jang, Larry Sheldon, San Jose, CA, United States Leung, Danton Kai-Yu, Los Altos, CA, United States Rocco, Richard Michele, Sunnyvale, CA, United States Platshon, Mark Charles, Menlo Park, CA, United States Idexx Laboratories, Inc., Westbrook, ME, United States

PATENT ASSIGNEE(S): Idexx Laboratorie

(U.S. corporation)

NUMBER KIND DATE

PATENT INFORMATION: US 5976896 19991102 APPLICATION INFO.: US 1996-688043 19960729 (8)

RELATED APPLN. INFO.: Continuation-in-part of Ser. No. US 1994-254302, filed

on 6 Jun 1994, now patented, Pat. No. US 5624850

DOCUMENT TYPE: Utility

Chin, Christopher L. PRIMARY EXAMINER: Graser, Jennifer ASSISTANT EXAMINER: Lyon & Lyon LLP LEGAL REPRESENTATIVE:

NUMBER OF CLAIMS: 11 EXEMPLARY CLAIM:

NUMBER OF DRAWINGS: 29 Drawing Figure(s); 24 Drawing Page(s)

2755 LINE COUNT:

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

A fluorescent immunoassay employing the interior surface of a capillary tube is provided. Devices to permit immunoassays using one or more capillary tubes, an apparatus for use with the devices, and a process for screening for analyte in a sample using the devices and apparatus are also provided. Samples suspected of containing analyte are added to a disposable self-contained sample tray containing one or more sample wells, mixed with a reagent, drawn into one or more spaced-apart capillary tubes held within a disposable cartridge connected to an analytical apparatus, reacted with a binding member on the surface of the capillary tube, washed to stop the reaction, and dried by the apparatus. The capillary tube is then exposed to a signal generation device to create a fluorescence signal that is detected using a signal detector. The apparatus determines the presence of the analyte and optionally determines the amount of analyte present in the sample, and presents the results to the operator.

L11 ANSWER 23 OF 40 USPATFULL

1999:120765 USPATFULL ACCESSION NUMBER:

Apparatus for laser alloying induced improvement of TITLE:

surfaces

McCay, Thurman Dwayne, Winchester, TN, United States INVENTOR(S):

McCay, Mary Helen, Winchester, TN, United States Dahotre, Narendra B., Tullahoma, TN, United States The University of Tennessee Research Corporation,

PATENT ASSIGNEE(S):

Knoxville, TN, United States (U.S. corporation)

NUMBER KIND DATE \_\_\_\_\_\_

US 5961861 US 1997-932013 19991005 APPLICATION INFO.: 19970917 (8)

RELATED APPLN. INFO.: Division of Ser. No. US 1996-587553, filed on 17 Jan

1996, now abandoned

DOCUMENT TYPE: Utility

PRIMARY EXAMINER: Utility
PRIMARY EXAMINER: Evans, Geoffrey S.

LEGAL REPRESENTATIVE: Rosenblatt & Redano P.C.

NUMBER OF CLAIMS: 20 EXEMPLARY CLAIM:

7 Drawing Figure(s); 3 Drawing Page(s) NUMBER OF DRAWINGS:

907 LINE COUNT:

A feedback control system for performing and controlling the AB

laser alloying of a workpiece. The apparatus includes a

laser beam delivery system, a movement system capable of causing relative movement between a laser beam and a workpiece being irradiated by the laser beam, a precursor application system capable of applying a precursor at a desired rate to the surface of a moving workpiece, and a control system capable of receiving input

signals indicative of one or more measured process parameters, processing those signals, and transmitting a control signal capable of controlling the laser beam delivery system, movement system, and/or precursor application system. Other embodiments of the invention utilize a variety of process parameter measuring devices in conjunction with the control system. These devices include, but are not limited to, temperature transducers, infrared detectors, and emission spectra measuring devices.

1999:99591 USPATFULL ACCESSION NUMBER:

TITLE: High throughput screening assay systems in microscale

fluidic devices

Parce, John Wallace, Palo Alto, CA, United States INVENTOR(S):

Kopf-Sill, Anne R., Portola Valley, CA, United States

Bousse, Luc J., Menlo Park, CA, United States

PATENT ASSIGNEE(S): Caliper Technologies Corporation, Palo Alto, CA,

United

States (U.S. corporation)

	NUMBER	KIND	DATE	
PATENT INFORMATION:	US 5942443		19990824	
APPLICATION INFO.:	US 1996-671987		19960628	(8)

NUMBER DATE \_\_\_\_\_\_

US 1996-15498 19960416 (60) PRIORITY INFORMATION:

DOCUMENT TYPE: Utility
PRIMARY EXAMINER: Housel, James C.
ASSISTANT EXAMINER: Portner, Ginny Allen

LEGAL REPRESENTATIVE: Townsend and Townsend and Crew, Murphy, Matthew B., Quine, Jonathan Alan

NUMBER OF CLAIMS: 71 EXEMPLARY CLAIM: 1

NUMBER OF DRAWINGS: 17 Drawing Figure(s); 14 Drawing Page(s)

1730 LINE COUNT:

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

Microfluidic devices and methods that are useful for performing high-throughput screening assays. In particular, the devices and methods

of the invention are useful in screening large numbers of different compounds for their effects on a variety of chemical, and preferably, biochemical systems.

L11 ANSWER 25 OF 40 USPATFULL

ACCESSION NUMBER: 1999:67276 USPATFULL

Two-photon upconverting dyes and applications TITLE:

Prasad, Paras N., Williamsville, NY, United States INVENTOR(S): Bhawalkar, Jayant D., Tonawanda, NY, United States

He, Guang S., Williamsville, NY, United States Zhao, Chan F., San Diego, CA, United States

Gvishi, Raz, K. Tiron, Israel

Ruland, Gary E., Grand Island, NY, United States Zieba, Jaroslaw, Santa Rosa, CA, United States Cheng, Ping Chin, Williamsville, NY, United States

Pan, Shan Jen, Amherst, NY, United States

The Research Foundation of State university of New PATENT ASSIGNEE(S):

York, Amherst, NY, United States (U.S. corporation)

NUMBER KIND DATE \_\_\_\_\_ US 5912257 US 1996-712143 19990615 PATENT INFORMATION:

19960905 (8) APPLICATION INFO.:

NUMBER DATE \_\_\_\_\_ US 1995-3296 19950906 (60) US 1995-5924 19951027 (60) US 1995-10330 19951215 (60) PRIORITY INFORMATION:

Utility DOCUMENT TYPE:

PRIMARY EXAMINER: Utility
Davis, Zinna Northington

LEGAL REPRESENTATIVE: Nixon, Hargrave, Devans & Doyle LLP

NUMBER OF CLAIMS: 80 EXEMPLARY CLAIM:

NUMBER OF DRAWINGS: 34 Drawing Figure(s); 34 Drawing Page(s)

6013 LINE COUNT:

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

Styryl dyes and compositions Which exhibit superior two-photon absorption cross-sections and are useful in two-photon pumped cavity lasing, two-photon pumped upconversion lasing, optical power limiting, optical power stabilization, optical signal reshaping, and infrared

beam

detection and indication are disclosed. Also disclosed are multiphasic nanostructured composites which include a glass having pores, an optically active coating material on the pore surface, and a polymeric material in the pores. These composites are useful in producing multifunctional optical materials, such as broadly tunable lasers. Methods for killing cells and viruses using a photosensitizer and a two-photon upconverting dye are also described. These methods are especially useful to kill cells and viruses in biological materials, such as in photodynamic therapy of tumors and cancers or blood purification protocols. Media and methods for recording data in a three-dimensional matrix which includes a plurality of dye molecules is also described. The data storage methods and media have approximately 10.sup.12 volume elements per square centimeter, and each of the volume elements can store a single bit, digital information, or analog information. The data storage methods and media of the present

invention

are particularly useful for storing or archiving a series of two-dimensional black and white or color images, such as frames of a movie.

L11 ANSWER 26 OF 40 USPATFULL

ACCESSION NUMBER: 1999:15786 USPATFULL

TITLE:

Raman optrode processes and devices for detection of

chemicals and microorganisms

INVENTOR(S):

Grow, Ann E., 5882 Highplace Dr., San Diego, CA,

United

States 92120

NUMBER KIND DATE \_\_\_\_\_ US 5866430 19990202 US 1997-864015 19970527 PATENT INFORMATION: 19970527 (8)

> NUMBER DATE \_\_\_\_\_

PRIORITY INFORMATION: US 1996-19742 19960613 (60)

APPLICATION INFO.:

DOCUMENT TYPE: Utility
PRIMARY EXAMINER: Weber, John P.

LEGAL REPRESENTATIVE: Beehler & Pavitt

NUMBER OF CLAIMS: 24 1 EXEMPLARY CLAIM:

NUMBER OF DRAWINGS: 20 Drawing Figure(s); 8 Drawing Page(s)

3934 LINE COUNT:

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

A methodology and devices for detecting or monitoring or identifying AΒ chemical or microbial analytes are described. The methodology comprises four basic steps: (1) The gas or liquid medium to be monitored or analyzed is brought into contact with a bioconcentrator which is used

to

bind with or collect and concentrate one or more analytes. (2) The bioconcentrator-analyte complex is then exposed to radiation of one or more predetermined wavelengths to produce Raman scattering spectral bands. (3) At least a portion of the Raman spectral bands are collected and processed by a Raman spectrometer to convert the same into an electrical signal. And (4) the electrical signal is processed to detect and identify, qualitatively and/or quantitatively, the analyte(s). The methodology of this invention may also comprise Raman reactive capacity analysis of the bioconcentrator itself, simultaneously with or independently from the detection of the analyte, to determine the

potential ability of the bioconcentrator to complex with analytes; the results of this latter analysis may be used to affect or alter or

modify

the methodology involved in detection and analysis of the analytes.

Also

the invention is accomplished by a Raman Optrode comprising: a bioconcentrator capable of binding with the analyte(s); a mechanism or procedure or device for bringing the gas or liquid sample into contact with the bioconcentrator; a light source suitable for generating Raman scattering; a Raman spectrometer capable of collecting and processing the Raman scattering spectral information and translating it into an electrical signal; and electronic hardware and software for analyzing the electrical signal and translating the signal into information on

the

presence, identity and/or quantity of the bound analytes. Various forms of bioconcentrators are described, as well as a variety of analytes which may be detected, monitored, or identified by this invention, and

variety of devices which can be fabricated based on this invention.

L11 ANSWER 27 OF 40 USPATFULL

ACCESSION NUMBER:

1998:138641 USPATFULL

TITLE:

Methods for measuring telomere length

INVENTOR(S):

Kozlowski, Michael R., Palo Alto, CA, United States

Prowse, Karen R., Groningen, Netherlands Wang, Sy-Shi, Burlingame, CA, United States Wong, Sharon, San Jose, CA, United States Kim, Nam Woo, San Jose, CA, United States

Allsopp, Richard, Menlo Park, CA, United States

PATENT ASSIGNEE(S):

Geron Corporation, Menlo Park, CA, United States (U.S.

corporation)

NUMBER KIND DATE \_\_\_\_\_\_ US 1996-660402 Continuation US 5834193 19981110

PATENT INFORMATION: APPLICATION INFO.:

19960607 (8)

RELATED APPLN. INFO.:

Continuation-in-part of Ser. No. US 1995-479916, filed

on 7 Jun 1995, now abandoned

DOCUMENT TYPE:

Utility

PRIMARY EXAMINER:

Zitomer, Stephanie W.

LEGAL REPRESENTATIVE: Kaster, Kevin R., Stracker, Elaine C.

NUMBER OF CLAIMS: 9

EXEMPLARY CLAIM:

4 Drawing Figure(s); 4 Drawing Page(s)

NUMBER OF DRAWINGS: LINE COUNT:

1906

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

Methods and compositions for the measurement of telomere length have AB application in medical diagnostic, prognostic, and therapeutic procedures. The methods for measuring telomere length include primer extension-based methods and probe-based methods. The primer extension methods involve elongation of telomeric, linker, and/or subtelomeric based primers under conditions such that the telomere serves as a template for primer extension and that the resultant primer extension products can be compared to standards of known length to provide a measure of telomere length. The probe based methods allow for telomere length measurements using DNA from lysed or whole cells and involve hybridizing an excess of probe to all telomeric repeat sequences in the telomere, measuring the amount of bound probe, and correlating the amount of bound probe measured with telomere length.

L11 ANSWER 28 OF 40 USPATFULL

ACCESSION NUMBER:

1998:131533 USPATFULL

TITLE:

Nucleic acid detection with energy transfer

INVENTOR(S):

Sammes, Peter George, Farnham Royal, United Kingdom

Garman, Andrew John, Chester, United Kingdom

PATENT ASSIGNEE(S):

Zeneca Limited, London, England (non-U.S. corporation)

NUMBER KIND DATE -----US 5827653 PATENT INFORMATION: 19981027 WO 9508642 19950330 APPLICATION INFO.: US 1996-619724 19960520 (8) WO 1994-GB2068 19940923 19960520 PCT 371 date 19960520 PCT 102(e) date

NUMBER DATE -----GB 1993-19826 19930923 GB 1994-12106 19940616 PRIORITY INFORMATION:

Utility DOCUMENT TYPE:

Sisson, Bradley L.

PRIMARY EXAMINER: LEGAL REPRESENTATIVE: Cushman Darby & Cushman Intellectual Property Group of

Pillsbury Madison & Sutro, LLP

NUMBER OF CLAIMS: EXEMPLARY CLAIM:

NUMBER OF DRAWINGS: 6 Drawing Figure(s); 3 Drawing Page(s)

LINE COUNT: 1807

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

A method for the detection of a nucleic acid analyte by complementary AB probe hybridisation and formation of a chelated lanthanide complex which, upon irradiation by light, results in a characteristic delayed luminescence emission.

L11 ANSWER 29 OF 40 USPATFULL

ACCESSION NUMBER: 97:70876 USPATFULL

TITLE: Luminescent lanthanide chelates and methods of use

INVENTOR(S): Selvin, Paul R., Berkeley, CA, United States Hearst, John, Berkeley, CA, United States

PATENT ASSIGNEE(S): The Regents of the University of California, Oakland,

CA, United States (U.S. corporation)

NUMBER KIND DATE -----PATENT INFORMATION: US 5656433 19970812 APPLICATION INFO.: US 1996-762288 19961209 (8)

Division of Ser. No. US 1994-269162, filed on 29 Jun RELATED APPLN. INFO.:

1994, now patented, Pat. No. US 5622821

DOCUMENT TYPE: Utility

PRIMARY EXAMINER: Marschel, Ardin H.
LEGAL REPRESENTATIVE: Osman, Richard Aron

NUMBER OF CLAIMS: 20 EXEMPLARY CLAIM:

NUMBER OF DRAWINGS: 6 Drawing Figure(s); 4 Drawing Page(s)

LINE COUNT: 1243

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

The invention provides lanthanide chelates capable of intense luminescence. The celates comprise a lanthanide chelator covalently joined to a coumarin-like or quinolone-like sensitizer. Exemplary sensitizers include 2- or 4-quinolones, 2- or 4-coumarins, or

derivatives thereof e.g. carbostyril 124

(7-amino-4-methyl-2-quinolone),

coumarin 120 (7-amino-4-methyl-2-coumarin), coumarin 124 (7-amino-4-(trifluoromethyl)-2-coumarin), aminomethyltrimethylpsoralen,

The chelates form high affinity complexes with lanthanides, such as terbium or europium, through chelator groups, such as DTPA. The chelates

may be coupled to a wide variety of compounds to create specific labels,

probes, diagnostic and/or therapeutic reagents, etc. The chelates find

particular use in resonance energy transfer between chelate-lanthanide complexes and another luminescent agent, often a fluorescent non-metal based resonance energy acceptor. The methods provide useful information about the structure, conformation, relative location and/or

interactions

of macromolecules.

L11 ANSWER 30 OF 40 USPATFULL

ACCESSION NUMBER:

97:51868 USPATFULL

TITLE:

Luminescent lanthanide chelates and methods of use

INVENTOR(S):

Selvin, Paul R., Berkeley, CA, United States Hearst, John, Berkeley, CA, United States

PATENT ASSIGNEE(S):

The Regents of the University of California, Oakland,

CA, United States (U.S. corporation)

NUMBER KIND DATE \_\_\_\_\_\_ PATENT INFORMATION: US 1996-762598 19970617 APPLICATION INFO.:

19961209 (8)

RELATED APPLN. INFO.:

Division of Ser. No. US 1994-269162, filed on 29 Jun

1994, now patented, Pat. No. US 5622821

DOCUMENT TYPE:

Utility

PRIMARY EXAMINER: LEGAL REPRESENTATIVE:

Marschel, Ardin H. Osman, Richard Aron

NUMBER OF CLAIMS:

20

EXEMPLARY CLAIM:

1

NUMBER OF DRAWINGS:

6 Drawing Figure(s); 4 Drawing Page(s)

LINE COUNT: 1215

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

The invention provides lanthanide chelates capable of intense luminescence. The celates comprise a lanthanide chelator covalently joined to a coumarin-like or quinolone-like sensitizer. Exemplary

sensitzers include 2- or 4-quinolones, 2- or 4-coumarins, or

derivatives

thereof e.g. carbostyril 124 (7-amino-4-methyl-2-quinolone), coumarin 120 (7-amino-4-methyl-2-coumarin), coumarin 124 (7-amino-4-(trifluoromethyl)-2-coumarin), aminomethyltrimethylpsoralen, etc.

The chelates form high affinity complexes with lanthanides, such as terbium or europium, through chelator groups, such as DTPA. The chelates

may be coupled to a wide variety of compounds to create specific labels,

probes, diagnostic and/or therapeutic reagents, etc. The chelates find particular use in resonance energy transfer between chelate-lanthanide complexes and another luminescent agent, often a fluorescent non-metal based resonance energy acceptor. The methods provide useful information about the structure, conformation, relative location and/or interactions

of macromolecules.

L11 ANSWER 31 OF 40 USPATFULL

ACCESSION NUMBER:

97:33616 USPATFULL

TITLE: INVENTOR(S):

Luminescent lanthanide chelates and methods of use

Selvin, Paul R., Berkeley, CA, United States Hearst, John, Berkeley, CA, United States

PATENT ASSIGNEE(S):

The Regents of the University of California, Oakland,

CA, United States (U.S. corporation)

NUMBER KIND DATE -----US 5622821 19970422 US 1994-269162 19940629 (8) PATENT INFORMATION: APPLICATION INFO.: DOCUMENT TYPE: Utility PRIMARY EXAMINER: Marschel, Ardin H.

LEGAL REPRESENTATIVE: Osman, Richard Aron

NUMBER OF CLAIMS: 18 EXEMPLARY CLAIM: 1

NUMBER OF DRAWINGS: 6 Drawing Figure(s); 4 Drawing Page(s)

LINE COUNT: 1254

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB The invention provides lanthanide chelates capable of intense luminescence. The celates comprise a lanthanide chelator covalently joined to a coumarin-like or quinolone-like sensitizer. Exemplary sensitzers include 2- or 4-quinolones, 2- or 4-coumarins, or

derivatives
thereof e.g. carbostyril 124 (7-amino-4-methyl-2-quinolone), coumarin
120 (7-amino-4-methyl-2-coumarin), coumarin 124 (7-amino-4-

(trifluoromethyl)-2-coumarin), aminomethyltrimethylpsoralen, etc.

The chelates form high affinity complexes with lanthanides, such as terbium or europium, through chelator groups, such as DTPA. The chelates

may be coupled to a wide variety of compounds to create specific labels,

probes, diagnostic and/or therapeutic reagents, etc. The chelates find particular use in resonance energy transfer between chelate-lanthanide complexes and another luminescent agent, often a fluorescent non-metal based resonance energy acceptor. The methods provide useful information about the structure, conformation, relative location and/or

interactions

of macromolecules.

L11 ANSWER 32 OF 40 USPATFULL

ACCESSION NUMBER: 96:60602 USPATFULL

TITLE: Fluorescent viability assay using cyclic-substituted

unsymmetrical cyanine dyes

INVENTOR(S): Millard, Paul J., Eugene, OR, United States

Roth, Bruce L., Corvallis, OR, United States Yue, Stephen T., Eugene, OR, United States Haugland, Richard P., Eugene, OR, United States

PATENT ASSIGNEE(S): Molecular Probes, Inc., Eugene, OR, United States

(U.S.

corporation)

RELATED APPLN. INFO.: Continuation of Ser. No. US 1993-90890, filed on 12

Jul

1993, now patented, Pat. No. US 5436134 And Ser. No.

US

is

1993-146328, filed on 1 Nov 1993, each which is a continuation-in-part of Ser. No. US 1993-47683, filed on 13 Apr 1993, now abandoned

DOCUMENT TYPE: Utility
PRIMARY EXAMINER: Kight, John

ASSISTANT EXAMINER: Leary, Louise N.

LEGAL REPRESENTATIVE: Helfenstein, Allegra J., Skaugset, Anton E.

NUMBER OF CLAIMS: 20 EXEMPLARY CLAIM: 1

NUMBER OF DRAWINGS: 5 Drawing Figure(s); 4 Drawing Page(s)

LINE COUNT: 1908

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB The invention relates to a method of analyzing the viability of a sample

of cells using an aqueous solution comprising two fluorescent dyes. Dye I has the formula: ##STR1## where R.sup.2 is C.sub.1-6 alkyl; Z.sup.-

a biologically compatible counterion;

X is O; S; Se; or NR.sup.15, where R.sup.15 is H or C.sub.1-6 alkyl; or CR.sup.16 R.sup.17, where R.sup.16 and R.sup.17, which may be the same or different, are independently H or C.sub.1-6 alkyl, or the carbons of R.sup.16 and R.sup.17 taken in combination complete a five or six membered saturated ring; and the benzazolium is optionally further substituted;

n=0, 1, or 2;

Y is --CR.sup.3 .dbd.CR.sup.4 --; p and m=0 or 1, such that p+m=1;

R.sup.5 is a C.sub.1-6 alkyl, C.sub.1-6 alkenyl, C.sub.1-6 polyalkenyl, C.sub.1-6 alkynyl or C.sub.1-6 polyalkynyl group; or R.sup.5 is an OMEGA;

R.sup.3, R.sup.4, R.sup.6 and R.sup.7, which may be the same or different, are independently H; or a C.sub.1-6 alkyl, C.sub.1-6 alkenyl,

C.sub.1-6 polyalkenyl, C.sub.1-6 alkynyl or C.sub.1-6 polyalkynyl group;

or halogen; or --OR.sup.8, --SR.sup.8, --(NR.sup.8 R.sup.9), where R.sup.8 and R.sup.9, which may be the same or different, are independently H; or alkyl groups having 1-6 carbons; or 1-2 substituted or unsubstituted alicyclic, heteroalicyclic, aromatic, or heteroaromatic

rings, containing 1-4 heteroatoms, wherein the heteroatoms are O, N, or S; or R.sup.8 and R.sup.9 taken in combination are --(CH.sub.2).sub.2 --L--(CH.sub.2).sub.2 -- where L=--O--, --NR.sup.10 --, --CH.sub.2 --

or

a single bond where R.sup.10 is H or an alkyl group having 1-6 carbons; or --OSO.sub.2 R.sup.19 where R.sup.19 is C.sub.1-6 alkyl, or C.sub.1-6 perfluoroalkyl, or aryl; or an OMEGA; or R.sup.6 and R.sup.7, taken in combination are --(CH.sub.2).sub.v -- where v=3 or 4, or R.sup.6 and R.sup.7 form a fused aromatic ring that is optionally further substituted;

such that at least one of R.sup.3, R.sup.4, R.sup.5, R.sup.6 and R.sup.7, or a substituent on the aromatic ring formed by R.sup.6 and R.sup.7, is an OMEGA; where OMEGA is a cyclic substituent that is attached by a single bond.

Fluorescent Dye II selectively stains either viable or non-viable cells with a detectable fluorescent response that is different from the fluorescent response of Dye I. The stained cells are illuminated at a suitable absorption wavelength, and the fluorescent response is detected

to distinguish viable and non-viable cells based on the fluorescent response.

L11 ANSWER 33 OF 40 USPATFULL

ACCESSION NUMBER: 96:16883 USPATFULL

TITLE: Immunodiagnostic assay using liposomes carrying labels

thereof on outer liposome surface

INVENTOR(S): Carbonell, Ruben G., Cary, NC, United States

Kilpatrick, Peter K., Cary, NC, United States Jones, Matthew A., Raleigh, NC, United States Singh, Anup K., Raleigh, NC, United States

PATENT ASSIGNEE(S): North Carolina State University, Raleigh, NC, United

States (U.S. corporation)

NUMBER KIND DATE

PATENT INFORMATION: US 5494803 19960227
APPLICATION INFO.: US 1994-273280 19940711 (8)

RELATED APPLN. INFO.: Continuation of Ser. No. US 1991-795910, filed on 19

Nov 1991, now abandoned

DOCUMENT TYPE: Utility

PRIMARY EXAMINER: PRIMARY EXAMINER: Scheiner, Toni R. ASSISTANT EXAMINER: Parsons, Nancy J.

LEGAL REPRESENTATIVE: Bell, Seltzer, Park & Gibson

NUMBER OF CLAIMS: 18 EXEMPLARY CLAIM: 1

NUMBER OF DRAWINGS: 16 Drawing Figure(s); 14 Drawing Page(s)

1473 LINE COUNT:

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

Competitive and sandwich-type immunodiagnostic assays can be configured by use of liposomes carrying detectible markers (e.g., fluorophores) or catalysts thereof (e.g., enzymes) on the outer liposome surface. The liposome also contains at least one antigen or antibody allowing it to bind to a complementary, immobilized antibody or antigen on a support.

L11 ANSWER 34 OF 40 USPATFULL

94:95433 USPATFULL ACCESSION NUMBER:

Molecular analytical release tags and their use in TITLE:

chemical analysis

INVENTOR(S): Giese, Roger W., Quincy, MA, United States

Northeastern University, Boston, MA, United States PATENT ASSIGNEE(S):

(U.S. corporation)

NUMBER KIND DATE \_\_\_\_\_\_ US 5360819 19941101 US 1985-710318 19850311 (6) PATENT INFORMATION: APPLICATION INFO.:

Continuation-in-part of Ser. No. US 1982-344394, filed on 1 Feb 1982, now patented, Pat. No. US 4709016 RELATED APPLN. INFO.:

DOCUMENT TYPE: Utility PRIMARY EXAMINER: Killos, Paul J.

LEGAL REPRESENTATIVE: Weingarten, Schurgin, Gagnebin & Hayes

NUMBER OF CLAIMS: EXEMPLARY CLAIM: 1583 LINE COUNT:

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

A release tag reagent suitable for use in the chemical analysis of a AB

substance to be detected, which substance contains reactive groups, such as for, but not limited to gas phase detection

groups, which reagent comprises three covalently bonded groups: a

signal

group which on release provides a ketone signal compound to be detected,

a release group which may be cleaved to release the ketone signal group,

which release group contains, for example, a vic glycol or an olefin group and a reactivity group which is reactive with a reactive group of the substance to be detected.

L11 ANSWER 35 OF 40 USPATFULL

91:33116 USPATFULL ACCESSION NUMBER:

Immunoassay using optical interference detection TITLE:

INVENTOR(S): Nicoli, David F., 448 Mills Way, Goleta, CA, United

States 93117

Elings, Virgil B., 1155 Via Tranquilla, Santa Barbara,

CA, United States 93110

NUMBER KIND DATE \_\_\_\_\_\_\_\_\_\_ US 33581 US 4647544 19910430 PATENT INFORMATION: 19870303 US 1987-72699 US 1984-624460 (Original) 19870713 (7) APPLICATION INFO.:

19840625 (Original)

DOCUMENT TYPE: Reissue
PRIMARY EXAMINER: Nucker, Christine

LEGAL REPRESENTATIVE: Walker, William B., Terlizzi, Laura

NUMBER OF CLAIMS: 40 49 EXEMPLARY CLAIM:

NUMBER OF DRAWINGS: 8 Drawing Figure(s); 3 Drawing Page(s)

1674 LINE COUNT:

AB Apparatus and method for providing an optical detection of a binding reaction between a ligand and an antiligand, including, a pattern formed

by a spatial array of microscopic dimensions of antiligand material, ligand material interacting with the antiligand material to produce a binding reaction between the ligand and the antiligand in the pattern,

а

source of optical radiation including energy at at least one wavelength directed to the pattern at a particular incidence angle to produce scattering of the energy from the pattern in accordance with the binding

reaction and with a strong scattering intensity at one or more Bragg scattering angles, and at least one optical detector located relative

to

the pattern and aligned with a Bragg scattering angle to detect the strong scattering intensity at the Bragg scattering angle to produce a signal representative of the binding reaction.

L11 ANSWER 36 OF 40 USPATFULL

89:78671 USPATFULL ACCESSION NUMBER:

Analyte detection by means of energy transfer TITLE:

Stavrianopoulos, Jannis, New York, NY, United States INVENTOR(S):

Rabbani, Elazar, New York, NY, United States Abrams, Samuel B., New York, NY, United States Wetmur, James G., Scardsdale, NY, United States

Enzo Biochem, Inc., New York, NY, United States (U.S. PATENT ASSIGNEE(S):

corporation)

NUMBER KIND DATE

US 4868103 US 1986-831250 APPLICATION INFO.:
DOCUMENT TYPE: 19890919 19860219 (6)

Utility

PRIMARY EXAMINER: ASSISTANT EXAMINER: Brown, Johnnie R. Jay, Jeremy M.

LEGAL REPRESENTATIVE: Mosoff, Serle I., Tzagoloff, Helen

NUMBER OF CLAIMS: 34 EXEMPLARY CLAIM:

NUMBER OF DRAWINGS: 7 Drawing Figure(s); 2 Drawing Page(s)

LINE COUNT: 1656

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

A method is disclosed to detect the presence of an analyte. The method involves forming a complex comprising the analyte and a binding entity. The binding entity comprises a first partner of an energy transfer system. The complex is then contacted with a reporting entity to form a unit. The reporting entity comprises a second partner of the energy transfer system. The first partner and the second partner are within Furster's radius of each other in the formed unit. The unit is irradiated with energy which can only be absorbed by one of said partners, namely, the energy donor, which then emits fluorescent

energy.

Some of this energy is absorbed by the other of said partners, namely, the energy acceptor, which also emits fluorescent energy. However, the fluorescent energy of the energy acceptor is of longer wavelength and

in

addition may be of substantially greater duration than the fluorescent energy of the energy donor. The detection of fluorescence at the longer wavelength or after a given time interval verifies the presence of the analyte.

L11 ANSWER 37 OF 40 USPATFULL

ACCESSION NUMBER: 89:30047 USPATFULL TITLE: Lifetime-resolved assay procedures

INVENTOR(S): Morrison, Larry E., Lisle, IL, United States

PATENT ASSIGNEE(S): Amoco Corporation, Chicago, IL, United States (U.S.

corporation)

NUMBER KIND DATE -----PATENT INFORMATION: 19890418

APPLICATION INFO.:

US 4822733 US 1985-738560

19850528 (6)

DOCUMENT TYPE:

Utility

PRIMARY EXAMINER:

ASSISTANT EXAMINER: LEGAL REPRESENTATIVE: Warden, Robert J. Benson, Robert Janiuk, Anthony J., Magidson, William H., Medhurst,

Ralph C.

NUMBER OF CLAIMS:

67 1

EXEMPLARY CLAÌM:

NUMBER OF DRAWINGS:

7 Drawing Figure(s); 4 Drawing Page(s)

LINE COUNT:

1636

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB Improved luminescent lifetime-resolved association assay techniques for detection of analytes in samples using two photophore-labelled probes, the photophores of which have different emissive lifetimes. One of the photophores is excitable by a modulated energy source to an excited state from which energy may be transferred to the other photophore when in close poximity thereto resulting in excitation and emission of the other photophore. Methods according to the invention involve

associating

the first photophore-labelled probe with the analyte and associating the

second photophore-labelled probe with the analyte or first probe in a reaction mixture bringing the photophores in sufficient proximity to allow energy transfer to occur. The reaction mixture is formed, excited by the modulated energy source and monitored for emission of the photophore excited by energy transfer at a time beyond the emissive lifetime of the shorter-lived photophore.

L11 ANSWER 38 OF 40 USPATFULL

ACCESSION NUMBER:

89:12832 USPATFULL

TITLE:

Ligand-receptor assays employing squarate dye

compositions

INVENTOR(S):

Berger, Jr., Donald E., San Jose, CA, United States Tarnowski, Thomas L., So. San Francisco, CA, United

Ullman, Edwin F., Atherton, CA, United States

PATENT ASSIGNEE(S):

Syntex (U.S.A.) Inc., Palo Alto, CA, United States

(U.S. corporation)

NUMBER KIND DATE ----- ---- -----US 4806488 19890221 US 1985-773401 19850906 PATENT INFORMATION: 19850906 (6) APPLICATION INFO.:

DOCUMENT TYPE:

Utility

Warden, Robert J. Wieder, Stephen C. PRIMARY EXAMINER: ASSISTANT EXAMINER: LEGAL REPRESENTATIVE: Leitereg, Theodore J.

NUMBER OF CLAIMS: EXEMPLARY CLAIM: LINE COUNT:

82 44 1497

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

Novel assays for ligands and receptors employing novel compounds that are conjugates of squarates dyes and members of a specific binding pair (sbp) are disclosed. The sbp members are selected from the group consisting of ligand and its complementary receptor. The sbp member is covalently or non-covalently bound to the squarate dye, which usually has an absorption maximum greater than 600 nanometers. The novel conjugates are employed in assays for determining the presence or

amount.

of an sbp member analyte in a sample suspected of containing such analyte. Kits comprising such novel conjugates are also disclosed.

L11 ANSWER 39 OF 40 USPATFULL

ACCESSION NUMBER: 87:18702 USPATFULL

TITLE: Method of chemical analysis employing molecular

release

tag compounds

Giese, Roger W., 56 Oakland Ave., Quincy, MA, United INVENTOR(S):

States 02170

NUMBER KIND DATE \_\_\_\_\_\_\_\_\_\_ PATENT INFORMATION: US 4650750 US 1984-591262 19870317

APPLICATION INFO.: 19840319 (6)

RELATED APPLN. INFO.: Division of Ser. No. US 1982-344394, filed on 1 Feb

1982

DOCUMENT TYPE: Utility

Wiseman, Thomas G. PRIMARY EXAMINER: PRIMARY EXAMINER: Wiseman, Thomas G. ASSISTANT EXAMINER: Teskin, Robin Lyn

LEGAL REPRESENTATIVE: Weingarten, Schurgin, Gagnebin & Hayes

NUMBER OF CLAIMS: EXEMPLARY CLAIM: 1 LINE COUNT: 582

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

A method of chemical analysis is disclosed which employs a release tag

compound of general formula

Rx--Re--S

in which Rx is a reactivity group capable of forming a covalent bond with another molecule, Re is a release group capable of being cleaved, and S is a signal group. The release tag is covalently bonded to a substance of interest which is to be determined in the course of a chemical analysis for an analyte, release group Re is cleaved at an appropriate point in the analytical procedure, and signal group S is determined, thereby determining the substance of interest. Where the substance of interest is the analyte, determination of S determines the analyte. Where the substance of interest is not the analyte but is related to the analyte concentration, determination of S allows

indirect

determination of analyte.

L11 ANSWER 40 OF 40 USPATFULL

ACCESSION NUMBER: 87:15235 USPATFULL

TITLE: Immunoassay using optical interference detection Nicoli, David F., 448 Mills Way, Goleta, CA, United INVENTOR(S):

States 93017

Elings, Virgil B., 1155 Via Tranquila, Santa Barbara,

CA, United States 93110

NUMBER KIND DATE

US 4647544 US 1984-624460 PATENT INFORMATION: 19870303 APPLICATION INFO.: 19840625 (6)

Utility DOCUMENT TYPE:

Nucker, Christine M. PRIMARY EXAMINER: ASSISTANT EXAMINER: Wieder, Stephen C.

LEGAL REPRESENTATIVE: Schwartz, Charles H., Roston, Ellsworth R.

NUMBER OF CLAIMS: 48 EXEMPLARY CLAIM:

8 Drawing Figure(s); 3 Drawing Page(s) NUMBER OF DRAWINGS:

1551 LINE COUNT:

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

Apparatus and method for providing an optical detection of a binding reaction between a ligand and an antiligand, including, a pattern

formed

by a spatial array of microscopic dimensions of antiligand material, ligand material interacting with the antiligand material to produce a binding reaction between the ligand and the antiligand in the pattern,

source of optical radiation including energy at at least one wavelength directed to the pattern at a particular incidence angle to produce scattering of the energy from the pattern in accordance with the binding

a

to

reaction and with a strong scattering intensity at one or more  ${\tt Bragg}$  scattering angles, and at least one optical detector located relative

the pattern and aligned with a Bragg scattering angle to detect the strong scattering intensity at the Bragg scattering angle to produce a signal representative of the binding reaction.